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201-16171B

I U C L I D

Data Set

Existing Chemical	: ID: 60-29-7
CAS No.	: 60-29-7
EINECS Name	: diethyl ether
EC No.	: 200-467-2
TSCA Name	: Ethane, 1,1'-oxybis-
Molecular Formula	: C ₄ H ₁₀ O

Producer related part	
Company	: Diethyl Ether Producers Association
Creation date	: 19.05.2005

Substance related part	
Company	: Diethyl Ether Producers Association
Creation date	: 19.05.2005

Status	:
Memo	: with DME

Printing date	: 10.01.2006
Revision date	:
Date of last update	: 10.01.2006

Number of pages	: 95
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Chapter (profile)	: Chapter: 1, 2, 3, 4, 5, 6, 7, 8, 10
Reliability (profile)	: Reliability: without reliability, 1, 2, 3, 4
Flags (profile)	: Flags: without flag, confidential, non confidential, WGK (DE), TA-Luft (DE), Material Safety Dataset, Risk Assessment, Directive 67/548/EEC, SIDS

1. General Information

Id 60-29-7

Date 10.01.2006

1.0.1 APPLICANT AND COMPANY INFORMATION

Type :
Name : Equistar Chemicals, LP (A Lyondell Company)
Contact person :
Date :
Street : One Houston Center, Suite 700, 1221 McKinney Street
Town : Houston, TX 77010
Country : United States
Phone :
Telefax :
Telex :
Cedex :
Email :
Homepage :

09.01.2006

Type :
Name : Hercules Incorporated
Contact person :
Date :
Street : 1313 N. Market Street
Town : Wilmington, DE 19894
Country : United States
Phone :
Telefax :
Telex :
Cedex :
Email :
Homepage :

10.01.2006

Type :
Name : B.V. CONSOLCO
Contact person :
Date :
Street : De Ruyterkade 44
Town : 1012 AA Amsterdam
Country : Netherlands
Phone : 020-6221444
Telefax : 020-6254449
Telex : 12458
Cedex :
Email :
Homepage :

Source : EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
11.02.2000

Type :
Name : BASF AG
Contact person :
Date :
Street : Karl-Bosch-Str
Town : 67056 Ludwigshafen
Country : Germany
Phone :

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Telefax :
Telex :
Cedex :
Email :
Homepage :

Source : EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
11.02.2000

Type :
Name : BP Chemicals Ltd.
Contact person :
Date :
Street : 76, Buckingham Palace Road
Town : SW1 WOSU London
Country : United Kingdom
Phone :
Telefax :
Telex :
Cedex :
Email :
Homepage :

Source : EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
11.02.2000

Type :
Name : Huels AG
Contact person :
Date :
Street : Postfach
Town : D-45764 Marl
Country : Germany
Phone :
Telefax :
Telex :
Cedex :
Email :
Homepage :

Source : EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
11.02.2000

Type :
Name : Petrasol B.V.
Contact person :
Date :
Street : P.O.Box 222
Town : 4200 AE Gorinchem
Country : Netherlands
Phone : +31 183 630555
Telefax : +31 183 632272
Telex : 23602 petr nl
Cedex :
Email :
Homepage :

Source : EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
11.02.2000

Type :
Name : Sodes

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Contact person :
Date :
Street : 44 rue Jean-Goujon
Town : 75008 Paris
Country : France
Phone : 142561287
Telefax : 142257346
Telex : 651646
Cedex :
Email :
Homepage :

Source : EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
11.02.2000

1.0.2 LOCATION OF PRODUCTION SITE, IMPORTER OR FORMULATOR

1.0.3 IDENTITY OF RECIPIENTS

1.0.4 DETAILS ON CATEGORY/TEMPLATE

1.1.0 SUBSTANCE IDENTIFICATION

1.1.1 GENERAL SUBSTANCE INFORMATION

Purity type :
Substance type : organic
Physical status : liquid
Purity :
Colour :
Odour :

Source : EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
11.02.2000

1.1.2 SPECTRA

1.2 SYNONYMS AND TRADENAMES

1,1'-Oxybisethane

Source : BASF AG Ludwigshafen
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
29.08.1996

3-Oxapentane

Source : BASF AG Ludwigshafen
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
29.08.1996

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Anaesthetic ether

Source : BASF AG Ludwigshafen
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
29.08.1996

Anesthesia ether

Source : BASF AG Ludwigshafen
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
29.08.1996

Anesthetic ether

Source : BASF AG Ludwigshafen
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
29.08.1996

Diethyl ether

Source : BASF AG Ludwigshafen
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
29.08.1996

Diethyl oxide

Source : BASF AG Ludwigshafen
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
29.08.1996

Diethylether

Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
30.05.1994

Diethylether, Ethoxyethaan, Ether, Ethyloxyde, Diethyloxyde

Source : B.V. CONSOLCO Amsterdam
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
28.02.1997

Diethylether; ethoxyethane

Source : ISIS/RISKLINE release VI, 1997, Haskoning
Petrasol B.V. Gorinchem
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
04.05.1998

Diethyloxyd

Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
25.05.1994

Ethane, 1,1'-oxybis-

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
14.09.1993

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Ethane, 1,1'-oxybis- (9CI)

Source : BASF AG Ludwigshafen
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
29.08.1996

Ethane, 1,1'-oxybis-

Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
30.05.1994

Ether

Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
25.05.1994

Ether (6CI)

Source : BASF AG Ludwigshafen
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
29.08.1996

Ethoxyethan

Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
25.05.1994

Ethoxyethane

Source : BASF AG Ludwigshafen
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
29.08.1996

Ethyl ether (8CI)

Source : BASF AG Ludwigshafen
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
29.08.1996

Ethylether

Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
25.05.1994

Ethyloxid

Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
30.05.1994

Pronarcol

Source : BASF AG Ludwigshafen

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29.08.1996

EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

Sulfuric ether

Source

: Sodes Paris

EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

30.05.1994

1.3 IMPURITIES

1.4 ADDITIVES

1.5 TOTAL QUANTITY

Quantity

: 10000 - 50000 tonnes in

Source

: EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

11.02.2000

1.6.1 LABELLING

Labelling

: as in Directive 67/548/EEC

Specific limits

: no data

Symbols

: F+, Xn, ,

Nota

: , C,

R-Phrases

: (12) Extremely flammable
(19) May form explosive peroxides
(22) Harmful if swallowed
(66) Repeated exposure may cause skin dryness or cracking
(67) Vapours may cause drowsiness and dizziness

S-Phrases

: (2) Keep out of reach of children
(9) Keep container in a well-ventilated place
(16) Keep away from sources of ignition - No smoking
(29) Do not empty into drains
(33) Take precautionary measures against static discharges

Source

: EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

11.02.2000

1.6.2 CLASSIFICATION

Classified

: as in Directive 67/548/EEC

Class of danger

: corrosive

R-Phrases

: (22) Harmful if swallowed

Specific limits

:

Source

: EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

11.02.2000

Classified

: as in Directive 67/548/EEC

Class of danger

: extremely flammable

R-Phrases

: (12) Extremely flammable

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Date 10.01.2006

Specific limits :

Source : EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
11.02.2000

Classified : as in Directive 67/548/EEC
Class of danger :
R-Phrases : (67) Vapours may cause drowsiness and dizziness
Specific limits :

Source : EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
11.02.2000

Classified : as in Directive 67/548/EEC
Class of danger :
R-Phrases : (19) May form explosive peroxides
Specific limits :

Source : EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
11.02.2000

Classified : as in Directive 67/548/EEC
Class of danger :
R-Phrases : (66) Repeated exposure may cause skin dryness or cracking
Specific limits :

Source : EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
11.02.2000

1.6.3 PACKAGING

1.7 USE PATTERN

Type of use : type
Category : Non dispersive use

Source : EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
11.02.2000

Type of use : type
Category : Use in closed system

Source : EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
11.02.2000

Type of use : type
Category : Use resulting in inclusion into or onto matrix

Source : EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
11.02.2000

Type of use : type
Category : Wide dispersive use

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Source 11.02.2000	: EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
Type of use Category	: industrial : Basic industry: basic chemicals
Source 11.02.2000	: EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
Type of use Category	: industrial : Chemical industry: used in synthesis
Source 11.02.2000	: EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
Type of use Category	: industrial : Fuel industry
Source 11.02.2000	: EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
Type of use Category	: industrial : Personal and domestic use
Source 11.02.2000	: EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
Type of use Category	: industrial : Photographic industry
Source 11.02.2000	: EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
Type of use Category	: use : Cleaning/washing agents and disinfectants
Source 11.02.2000	: EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
Type of use Category	: use : Explosives
Source 11.02.2000	: EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
Type of use Category	: use : Fuel
Source 11.02.2000	: EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
Type of use Category	: use : Intermediates
Source 11.02.2000	: EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
Type of use Category	: use : Laboratory chemicals
Source 11.02.2000	: EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

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Type of use : use
Category : Pharmaceuticals

Source : EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
11.02.2000

Type of use : use
Category : Photochemicals

Source : EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
11.02.2000

Type of use : use
Category : Solvents

Source : EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
11.02.2000

1.7.1 DETAILED USE PATTERN

1.7.2 METHODS OF MANUFACTURE

1.8 REGULATORY MEASURES

1.8.1 OCCUPATIONAL EXPOSURE LIMIT VALUES

Type of limit : MAK (DE)
Limit value : 400 ml/m³
Short term exposure limit value
Limit value : 1600 ml/m³
Time schedule : 15 minute(s)
Frequency : 4 times

Country : Germany
Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
18.02.1997

Type of limit : MAK (DE)
Limit value : 1200 mg/m³
Short term exposure limit value
Limit value : 4800 mg/m³
Time schedule : 15 minute(s)
Frequency : 4 times

Country : Germany
Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
18.02.1997

Type of limit : OES (UK)
Limit value : 400 ml/m³
Short term exposure limit value
Limit value : 500 ml/m³

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Time schedule :
Frequency : times

Source : BP Chemicals Ltd. London
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
02.06.1994

Type of limit : other
Limit value : 1200 mg/m3
Short term exposure limit value
Limit value : 1500 mg/m3
Time schedule : 15 minute(s)
Frequency : times

Remark : Mean Exposure Limit Value (VME)
Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
30.05.1994

Type of limit : other
Limit value : 400 ml/m3
Short term exposure limit value
Limit value : 500 ml/m3
Time schedule : 15 minute(s)
Frequency : times

Remark : Mean Exposure Limit Value (VME)
Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
30.05.1994

Type of limit :
Limit value : 400 other

Remark : Opmerking: andere = ppm
Source : B.V. CONSOLCO Amsterdam
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
18.11.2003

1.8.2 ACCEPTABLE RESIDUES LEVELS

1.8.3 WATER POLLUTION

Classified by : KBwS (DE)
Labelled by : KBwS (DE)
Class of danger : 1 (weakly water polluting)

Country : Germany
Remark : Katalog-Nr. 80
Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
18.02.1997 (62)

1.8.4 MAJOR ACCIDENT HAZARDS

Legislation : Störfallverordnung (DE)
Substance listed : yes

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No. in Seveso directive :

Country : Germany
Remark : im Anhang IV genannt (Kat. 6; leichtentzündliche
Fluessigkeiten)
Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
18.02.1997 (62)

1.8.5 AIR POLLUTION

Classified by : TA-Luft (DE)
Labelled by : TA-Luft (DE)
Number : 3.1.7 (organic substances)
Class of danger : III
Country : Germany
Remark : Anhang E
Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
18.02.1997 (62)

1.8.6 LISTINGS E.G. CHEMICAL INVENTORIES

1.9.1 DEGRADATION/TRANSFORMATION PRODUCTS

1.9.2 COMPONENTS

1.10 SOURCE OF EXPOSURE

Remark : Diethylether is released into the atmosphere. Because of its
high vapor pressure and volatility, diethylether emissions
are expected to occur chiefly by means of exhaust resulting
during production and use.
In troposphere, the half life time of diethylether is
estimated at 43 hours (see RE:1).
Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
30.05.1994 (51)

Remark : Initial partitioning
Release into the Atmosphere
Because of its high vapor pressure and volatility, diethyl
ether emissions are expected to occur chiefly by means of
exhaust resulting during production and use.
Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
17.02.1997 (18)

Remark : Huels: Emissionserklaerung 1992
Release into the atmosphere on production site in 1992: 5000

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Source : kg/a
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
17.02.1997 (61)

1.11 ADDITIONAL REMARKS

1.12 LAST LITERATURE SEARCH

1.13 REVIEWS

2. Physico-Chemical Data

Id 60-29-7

Date 10.01.2006

2.1 MELTING POINT

Value : = -116.2 °C

Reliability : (2) valid with restrictions
Flag : Critical study for SIDS endpoint
14.11.2003 (53)

Value : = -116.3 °C

Decomposition : no, at °C

Sublimation : no

Method :

Year :

GLP :

Test substance :

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
17.02.1997 (55)

Value : -116 °C

Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
13.12.1993 (90)

2.2 BOILING POINT

Value : = 34.5 °C at 1013 hPa

Reliability : (2) valid with restrictions
Flag : Critical study for SIDS endpoint
14.11.2003 (53)

Value : 34 °C at 1013 hPa

Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
13.12.1993 (90)

Value : = 34.5 °C at 1013 hPa

Decomposition : no

Method :

Year :

GLP :

Test substance :

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
17.02.1997 (55)

2.3 DENSITY

Type : density

Value : = .7138 g/cm³ at 20 °C

2. Physico-Chemical Data

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Reliability : (2) valid with restrictions
Flag : Critical study for SIDS endpoint
14.11.2003 (53)

Type : density
Value : .71 g/cm³ at 20 °C
Method :
Year :
GLP : no
Test substance :

Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
13.12.1993 (90)

Type : density
Value : = .714 g/cm³ at 20 °C
Method :
Year :
GLP : no
Test substance :

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
17.02.1997 (55)

2.3.1 GRANULOMETRY

2.4 VAPOUR PRESSURE

Value : = 589 hPa at 20 °C

Reliability : (2) valid with restrictions
Flag : Critical study for SIDS endpoint
14.11.2003 (98)

Value : = 563 hPa at 20 °C

Result : Values at other temperatures:
0 degree C: 189 hPa
10 degree C: 389 hPa
30 degree C: 863 hPa
40 degree C: 1228 hPa
60 degree C: 2311 hPa
80 degree C: 3964 hPa
100 degree C: 6472 hPa

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
17.02.1997 (55)

Value : 587 hPa at 20 °C

Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
13.12.1993 (90)

Value : = 587 hPa at 20 °C

2. Physico-Chemical Data

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Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
17.02.1997 (62)

2.5 PARTITION COEFFICIENT

Partition coefficient :
Log pow : = .82 at 23 °C
pH value :
Method : OECD Guide-line 107 "Partition Coefficient (n-octanol/water), Flask-shaking Method"
Year : 1981
GLP : no
Test substance :

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
Reliability : (2) valid with restrictions
Flag : Critical study for SIDS endpoint
29.12.2003 (63)

Partition coefficient : octanol-water
Log pow : = .89 at °C
pH value :
Method : other (calculated): EPIWIN (v 3.11) KOWWIN Submodel (v 1.67)
Year : 2003
GLP :
Test substance :

Remark : The cited value is from the Experimental Database match in the model.
The calculated value was 1.05.
The EPIWIN model was run using the following measured physical chemical properties:
Water solubility (mg/L): 65000;
Vapor pressure (mm Hg): 442;
Log Kow (octanol-water): 0.82;
Boiling point (deg C): 34.50; and
Melting point (deg C): -116.20.

Reliability : (2) valid with restrictions
20.11.2003 (103)

Partition coefficient :
Log pow : .82 at 23 °C
pH value :
Method : OECD Guide-line 107 "Partition Coefficient (n-octanol/water), Flask-shaking Method"
Year :
GLP : no
Test substance :

Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
13.12.1993 (64)

Partition coefficient :
Log pow : .87 at °C
pH value :
Method : other (calculated): Leo, Hansch: Berechnung mit dem MedChem-Programm, Version 1989(POMONA89).
Year :

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GLP :
Test substance :

Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
14.09.1993

Partition coefficient :
Log pow : = .87 at °C
pH value :
Method : other (calculated): CLOGP3 Computer program according to Leo & Hansch
(MedChem, Version 1989)

Year :
GLP :
Test substance :

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
17.02.1997

2.6.1 SOLUBILITY IN DIFFERENT MEDIA

Solubility in : Water
Value : = 65 g/l at 20 °C
pH value :
concentration : at °C
Temperature effects :
Examine different pol. :
pKa : at 25 °C
Description :
Stable :

Result : Slightly water soluble at room temperature.
Reliability : (2) valid with restrictions
Flag : Critical study for SIDS endpoint
18.11.2003

(53)

Solubility in : Water
Value : 60 g/l at 25 °C
pH value :
concentration : at °C
Temperature effects :
Examine different pol. :
pKa : at 25 °C
Description :
Stable :
Deg. product :
Method : other: EPIWIN (v 3.11) WSKOWWIN Submodel (v 1.41)
Year : 2003
GLP :
Test substance :

Remark : The EPIWIN model was run using the following measured physical
chemical properties:
Water solubility (mg/L): 65000;
Vapor pressure (mm Hg): 442;
Log Kow (octanol-water): 0.82;
Boiling point (deg C): 34.50; and
Melting point (deg C): -116.20.

Result : Value represents experimental database value from model.

2. Physico-Chemical Data

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Date 10.01.2006

Reliability : The model estimated value was 30 g/l
21.11.2003 : (2) valid with restrictions (105)

Solubility in : Water
Value : 70 g/l at 20 °C
pH value : 7
concentration : at 20 °C
Temperature effects :
Examine different pol. :
pKa : at 25 °C
Description : of high solubility
Stable :
Deg. product :
Method :
Year :
GLP : no
Test substance :

Source : Sodes Paris
13.12.1993 EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (64)

Solubility in : Water
Value : = 70 g/l at 20 °C
pH value : 7
concentration : at 20 °C
Temperature effects :
Examine different pol. :
pKa : at 25 °C
Description : of high solubility
Stable :
Deg. product :
Method :
Year :
GLP : no
Test substance :

Source : Huels AG Marl
17.02.1997 EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (62)

Solubility in : Water
Value : = 65 g/l at 20 °C
pH value :
concentration : at °C
Temperature effects :
Examine different pol. :
pKa : at 25 °C
Description :
Stable :
Deg. product :
Method : other: no data
Year :
GLP :
Test substance :

Result : Values at other temperatures:
0 degree C: 117 g/l
10 degree C: 87 g/l
30 degree C: 52 g/l

2. Physico-Chemical Data

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Date 10.01.2006

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
17.02.1997 (55)

2.6.2 SURFACE TENSION

2.7 FLASH POINT

Value : = -45 °C
Type :
10.10.2003 (98)

Value : -40 °C
Type : closed cup
Method : other: DIN 51755
Year :
GLP : no
Test substance :
Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
13.12.1993 (64)

Value : = -40 °C
Type : closed cup
Method : other: DIN 51755
Year :
GLP : no
Test substance :
Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
17.02.1997 (55) (62)

2.8 AUTO FLAMMABILITY

Value : = 180 °C at
Method : other: DIN 51794
Year :
GLP :
Test substance :
Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
17.02.1997 (55) (62)

2.9 FLAMMABILITY

2.10 EXPLOSIVE PROPERTIES

2.11 OXIDIZING PROPERTIES**2.12 DISSOCIATION CONSTANT****2.13 VISCOSITY****2.14 ADDITIONAL REMARKS**

Memo : Explosive limits, peroxide formation

Remark : Explosive limits: lower limit 1.7 % v/v
upper limit 48 % v/v
Peroxides, which easily form in the presence of atmospheric oxygen, in particular under the influence of light, tend to explode when diethyl ether is distilled. Therefore, the presence of peroxide should always be tested before diethyl ether is utilized. Usually inhibitors are added.

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

17.02.1997

(62)

3. Environmental Fate and Pathways

Id 60-29-7

Date 10.01.2006

3.1.1 PHOTODEGRADATION

Type	:	air
Light source	:	
Light spectrum	:	nm
Relative intensity	:	based on intensity of sunlight
DIRECT PHOTOLYSIS		
Half-life t _{1/2}	:	= 9.8 hour(s)
Degradation	:	% after
Quantum yield	:	
INDIRECT PHOTOLYSIS		
Sensitizer	:	OH
Conc. of sensitizer	:	500000 molecule/cm ³
Rate constant	:	= .0000000000133 cm ³ /(molecule*sec)
Degradation	:	% after
Deg. product	:	
Method	:	other (measured): method not specified
Year	:	1987
GLP	:	no data
Test substance	:	no data
Remark	:	Half-life refers to 24-hour days. Photodegradation value was reported in this manuscript to compare to calculated data. The measured value (13.3E-12 cm ³ /molecule*sec) compared well with the calculated value (10.6E-12) that was reported in the manuscript.
Source	:	Huels AG Marl
Reliability	:	(2) valid with restrictions
Flag	:	Critical study for SIDS endpoint
29.12.2003		(7)
Type	:	other: EPIWIN (v 3.11) AOPWIN Submodel (v 1.91)
Light source	:	
Light spectrum	:	nm
Relative intensity	:	based on intensity of sunlight
DIRECT PHOTOLYSIS		
Half-life t _{1/2}	:	= 10.4 hour(s)
Degradation	:	% after
Quantum yield	:	
Deg. product	:	
Method	:	other (calculated): EPIWIN (v 3.11) AOPWIN Submodel (v 1.91)
Year	:	2003
GLP	:	
Test substance	:	
Remark	:	Overall OH rate constant = 12.3468 E-12 cm ³ /molecule-sec The EPIWIN model was run using the following measured physical chemical properties: Water solubility (mg/L): 65000; Vapor pressure (mm Hg): 442; Log Kow (octanol-water): 0.82; Boiling point (deg C): 34.50; and Melting point (deg C): -116.20.
Reliability	:	(2) valid with restrictions
29.12.2003		(101)

3. Environmental Fate and Pathways

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3.1.2 STABILITY IN WATER

Remark : Expert statement: Does not react with water; the only functionality other than carbon-carbon and carbon-hydrogen bonds is the ether linkage (C-O-C) which does not hydrolyze.

Reliability : (2) valid with restrictions

Flag : Critical study for SIDS endpoint

18.11.2003

3.1.3 STABILITY IN SOIL

3.2.1 MONITORING DATA

Type of measurement : background concentration

Media : surface water

Concentration :

Method :

Remark : Diethyl ether was found in 9 of 204 water samples from a nationwide study in the USA. Measurable concentrations in surface waters ranged from 0.003 mg/l (canal system on Lake Michigan) to 0.005 mg/l (Lake Michigan shore zone). Concentrations in sewage plant effluents varied from 0.001 to 0.01 mg/l.

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
17.02.1997 (108)

Type of measurement : concentration at contaminated site

Media : air

Concentration :

Method :

Remark : Young and Parker examined several different types of landfills during their research of gaseous components of various refuse landfills in Great Britain. Gaseous diethyl ether could only be detected in the ventilation gases of only one of the municipal refuse landfills at a concentration of < 20 mg/m3.

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
17.02.1997 (117)

Type of measurement : concentration at contaminated site

Media : air

Concentration :

Method :

Remark : In studies of landfill gases of two landfills in southern Germany, diethyl ether was detected but not quantified in the gas from a hazardous waste landfill but not in the gas from a municipal waste landfill.

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
17.02.1997 (66)

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- Type of measurement** : concentration at contaminated site
Media : ground water
Concentration :
Method :
- Remark** : Diethyl ether concentrations ranging from 0.002 to 1.5 mg/l were determined in the groundwater of a chemical landfill in the Netherlands which was openly operated without any groundwater protection measures during the period of 1960-1980, and where the disposed chemicals were regularly incinerated.
Accumulation of diethyl ether was observed in the deeper clay layer of the southeast sampling site, whereby a concentration of 150 mg/l was measured in the groundwater samples taken there. Individual measurements of lower layers, in which diethyl ether was mostly undetected, indicated that this substance was adsorbed stronger to clay than to the soil of the surrounding layers and that it was leached out much slower from the clay layer than the other ground layers; the flow of leachate was apparently obstructed in the clay layer.
- Source** : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
17.02.1997 (54)
- Type of measurement** : concentration at contaminated site
Media : ground water
Concentration :
Method :
- Remark** : Diethyl ether at concentrations near 0.0025 mg/l was found in 2 of 9 examined drinking water and groundwater samples from wells in the vicinity of a municipal and industrial landfill operated for 8 years in Delaware (USA). No diethyl ether could be detected in water samples from artesian wells which pump water for public use and which are located very close to this landfill (detection limit not reported).
- Source** : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
17.02.1997 (32)
- Type of measurement** : concentration at contaminated site
Media : ground water
Concentration :
Method :
- Remark** : In Gloucester, Canada, diethyl ether was found in groundwater below a landfill (municipal, partially with hazardous wastes) at concentrations of ≥ 5 mg/l (central area) and ≤ 0.1 mg/l (about 50 m away; Devlin & Gorman). Other authors (Patterson et al., Chaput et al.) reported > 10 mg/l in groundwater of the central area of this landfill and "undetectable" (detection limit not reported) 50-70 m away.
- Source** : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
17.02.1997 (21) (31) (83)
- Type of measurement** : other: city and national forest; potentially natural source
Media : air
Concentration :
Method :

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Remark : By a comparison of gaseous components of city air (Tuscaloosa, USA) and air from an unpopulated area (Talladega National Forest, USA), diethyl ether was detected but not quantified in both regions.

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (60)

17.02.1997

Type of measurement : other: natural source
Media : biota
Concentration :
Method :

Remark : Diethyl ether was found in gases transpired from the moss Polytrichum commune. The mechanism of formation and the quantity formed are not reported.

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (65)

17.02.1997

3.2.2 FIELD STUDIES

3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

Type : volatility
Media : water - air
Air : % (Fugacity Model Level I)
Water : % (Fugacity Model Level I)
Soil : % (Fugacity Model Level I)
Biota : % (Fugacity Model Level II/III)
Soil : % (Fugacity Model Level II/III)
Method : other: calculation from vapour pressure and water solubility
Year :

Remark : Henry's Law Constant was calculated from the data reported in the reference.

Result : Henry's Law Constant
- at 20 degree C: 64.20 Pa m3/mol
- at 0 degree C: 11.97 Pa m3/mol

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (55)

17.02.1997

3.3.2 DISTRIBUTION

Media : other: air (emissions to compartment = 1000 kg/hr)
Method : Calculation according Mackay, Level III
Year : 2003

Method : Equilibrium Concentration Model (EQC) Level III
Remark : The EPIWIN model was run using the following measured physical chemical properties:
Water Solubility (mg/L): 65000
Vapor Pressure (mm Hg) : 442
Log Kow (octanol-water): 0.82
Boiling Point (deg C) : 34.50

3. Environmental Fate and Pathways

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Date 10.01.2006

Result

Melting Point (deg C) : -116.20
: Concentration (%):
Air = 98.3
Water = 1.6
Soil ~ 0.1%
Sediment < 0.01

Level III Fugacity Model (Full-Output):

=====

Chem Name : Ethane, 1,1'-oxybis-
Molecular Wt: 74.12
Henry's LC : 0.00123 atm-m3/mole (Henry database)
Vapor Press : 442 mm Hg (user-entered)
Log Kow : 0.82 (user-entered)
Soil Koc : 2.71 (calc by model)

	Mass Amount (percent)	Half-Life (hr)	Emissions (kg/hr)
Air	98.2	19.6	1000
Water	1.64	360	0
Soil	0.121	360	0
Sediment	0.0029	1.44e+003	0

	Fugacity (atm)	Reaction (kg/hr)	Advection (kg/hr)	Reaction (percent)	Advection (percent)
Air	7.26e-011	779	220	77.9	22
Water	3.06e-011	0.709	0.369	0.0709	0.0369
Soil	6.69e-011	0.0524	0	0.00524	0
Sediment	2.53e-011	0.000312	1.3e-005	3.12e-005	1.3e-006

Persistence Time: 22.4 hr
Reaction Time: 28.8 hr
Advection Time: 102 hr
Percent Reacted: 77.9
Percent Advected: 22.1

Half-Lives (hr), (based upon Biowin (Ultimate) and Aopwin):

Air: 19.6
Water: 360
Soil: 360
Sediment: 1440
Biowin estimate: 3.027 (weeks)

Advection Times (hr):

Air: 100
Water: 1000
Sediment: 5e+004

Reliability**Flag**

30.11.2005

: (2) valid with restrictions
: Critical study for SIDS endpoint

(104)

Media**Method****Year**

: other: water (emissions to compartment = 1000 kg/hr)
: Calculation according Mackay, Level III
: 2003

Method**Remark**

: Equilibrium Concentration Model (EQC) Level III
: The EPIWIN model was run using the following measured physical chemical properties:
Water Solubility (mg/L): 65000
Vapor Pressure (mm Hg) : 442
Log Kow (octanol-water): 0.82
Boiling Point (deg C) : 34.50

3. Environmental Fate and Pathways

Id 60-29-7

Date 10.01.2006

Result

Melting Point (deg C) : -116.20
: Concentration (%)
Air = 4.4
Water = 95.4
Soil < 0.01
Sediment ~ 0.1

Level III Fugacity Model (Full-Output):

=====

Chem Name : Ethane, 1,1'-oxybis-
Molecular Wt: 74.12
Henry's LC : 0.00123 atm-m3/mole (Henry database)
Vapor Press : 442 mm Hg (user-entered)
Log Kow : 0.82 (user-entered)
Soil Koc : 2.71 (calc by model)

	Mass Amount (percent)	Half-Life (hr)	Emissions (kg/hr)
Air	4.38	19.6	0
Water	95.4	360	1000
Soil	0.00541	360	0
Sediment	0.168	1.44e+003	0

	Fugacity (atm)	Reaction (kg/hr)	Advection (kg/hr)	Reaction (percent)	Advection (percent)
Air	3.02e-011	324	91.6	32.4	9.16
Water	1.66e-008	384	200	38.4	20
Soil	2.78e-011	0.0218	0	0.00218	0
Sediment	1.37e-008	0.169	0.00704	0.0169	0.000704

Persistence Time: 209 hr
Reaction Time: 295 hr
Advection Time: 718 hr
Percent Reacted: 70.9
Percent Advected: 29.1

Half-Lives (hr), (based upon Biowin (Ultimate) and Aopwin):

Air: 19.6
Water: 360
Soil: 360
Sediment: 1440
Biowin estimate: 3.027 (weeks)

Advection Times (hr):

Air: 100
Water: 1000
Sediment: 5e+004

**Reliability
Flag**

30.11.2005

: (2) valid with restrictions
: Critical study for SIDS endpoint

(104)

**Media
Method
Year**

: air - biota - sediment(s) - soil - water
: Calculation according Mackay, Level I
:

Result

: Air: 95.621 %
Soil: 0.002 %
Water: 4.375 %
Sediment: 0.002 %
Biota: 0.000 %

Source

: Huels AG, Marl

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Test condition

Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
: Data used:
Molar mass: 74.12 g/mol
Log Pow: 0.82
Vapour pressure: 58700 Pa
Water solubility: 70.0 g/l

Equations used for additional data:
log Koc = 0.989 log Pow - 0.346

Volumes used:
Air: 6 000 000 000
Soil: 45 000
Water: 7 000 000
Sediment: 35 + 21 000
Biota: 7

14.11.2003

Media Method Year

: other: air - water - soil (emissions to compartments = 1000 kg/hr)
: Calculation according Mackay, Level III
: 2003

Method Remark

: Equilibrium Concentration Model (EQC) Level III
: The EPIWIN model was run using the following measured physical chemical properties:
Water Solubility (mg/L): 65000
Vapor Pressure (mm Hg) : 442
Log Kow (octanol-water): 0.82
Boiling Point (deg C) : 34.50
Melting Point (deg C) : -116.20

Result

: Concentration (%):
Air = 15.6
Water = 64.9
Soil = 19.4
Sediment < 1.0

Level III Fugacity Model (Full-Output):

=====

Chem Name : Ethane, 1,1'-oxybis-
Molecular Wt: 74.12
Henry's LC : 0.00123 atm-m3/mole (Henry database)
Vapor Press : 442 mm Hg (user-entered)
Log Kow : 0.82 (user-entered)
Soil Koc : 2.71 (calc by model)

	Mass Amount (percent)	Half-Life (hr)	Emissions (kg/hr)
Air	15.6	19.6	1000
Water	64.9	360	1000
Soil	19.4	360	1000
Sediment	0.114	1.44e+003	0

	Fugacity (atm)	Reaction (kg/hr)	Advection (kg/hr)	Reaction (percent)	Advection (percent)
Air	1.65e-010	1.77e+003	500	59	16.7
Water	1.73e-008	402	209	13.4	6.95
Soil	1.53e-007	120	0	3.99	0
Sediment	1.43e-008	0.177	0.00735	0.0059	0.000245

Persistence Time: 107 hr

Reaction Time: 140 hr

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Advection Time: 453 hr
Percent Reacted: 76.4
Percent Advected: 23.6

Half-Lives (hr), (based upon Biowin (Ultimate) and Aopwin):

Air: 19.6
Water: 360
Soil: 360
Sediment: 1440
Biowin estimate: 3.027 (weeks)

Advection Times (hr):

Air: 100
Water: 1000
Sediment: 5e+004

Reliability : (2) valid with restrictions
Flag : Critical study for SIDS endpoint
30.11.2005

(104)

Media : other: soil (emissions to compartment = 1000 kg/hr)
Method : Calculation according Mackay, Level III
Year : 2003

Method : Equilibrium Concentration Model (EQC) Level III
Remark : The EPIWIN model was run using the following measured physical chemical properties:

Water Solubility (mg/L): 65000
Vapor Pressure (mm Hg) : 442
Log Kow (octanol-water): 0.82
Boiling Point (deg C) : 34.50
Melting Point (deg C) : -116.20

Result : Concentration (%):

Air = 21
Water = 9.6
Soil = 69.4
Sediment < 0.1

Level III Fugacity Model (Full-Output):

=====

Chem Name : Ethane, 1,1'-oxybis-
Molecular Wt: 74.12
Henry's LC : 0.00123 atm-m3/mole (Henry database)
Vapor Press : 442 mm Hg (user-entered)
Log Kow : 0.82 (user-entered)
Soil Koc : 2.71 (calc by model)

	Mass Amount (percent)	Half-Life (hr)	Emissions (kg/hr)
Air	21	19.6	0
Water	9.57	360	0
Soil	69.4	360	1000
Sediment	0.0169	1.44e+003	0

	Fugacity (atm)	Reaction (kg/hr)	Advection (kg/hr)	Reaction (percent)	Advection (percent)
Air	6.21e-011	667	189	66.7	18.9
Water	7.12e-010	16.5	8.58	1.65	0.858
Soil	1.53e-007	120	0	12	0
Sediment	5.89e-010	0.00727	0.000302	0.000727	3.02e-005

Persistence Time: 89.6 hr
Reaction Time: 112 hr

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Advection Time: 455 hr

Percent Reacted: 80.3

Percent Advected: 19.7

Half-Lives (hr), (based upon Biowin (Ultimate) and Aopwin):

Air: 19.6

Water: 360

Soil: 360

Sediment: 1440

Biowin estimate: 3.027 (weeks)

Advection Times (hr):

Air: 100

Water: 1000

Sediment: 5e+004

Reliability : (2) valid with restrictions
Flag : Critical study for SIDS endpoint
30.11.2005

(104)

3.4 MODE OF DEGRADATION IN ACTUAL USE

3.5 BIODEGRADATION

Type : aerobic
Inoculum : activated sludge, non-adapted
Concentration : 100 mg/l related to
related to
Contact time : 14 day(s)
Degradation : 0 (±) % after 240 hour(s)
Result : under test conditions no biodegradation observed
Deg. product :
Method : other: similar to OECD 301C
Year : 1986
GLP : no data
Test substance : no data

Method : This study investigated the biological degradation of the test substance in a static electrolytic respirometer test for 14 days. Each culture flask containing 100 mg/L of diethyl ether in 300 ml of test solution and 1 ml JIS inorganic medium was inoculated with 30 mg/l non-acclimatized, activated sewage sludge and incubated at 20±1°C. The ThOD of diethyl ether was 2.59 g/g and the DOC was 0.65 g/g. The pH of the test solution was 7±1. The temperature was 20 ± 1 degrees C and the exposure period was 14 days. Measurements of biochemical oxygen demand (BOD) and removal of DOC were repeated 2-3 times.

Remark : In this test system, diethyl ether was not biodegradable after 240 h.
Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

Reliability : (2) valid with restrictions
Flag : Critical study for SIDS endpoint
18.11.2003

(107)

Type : aerobic
Inoculum : activated sludge, domestic, non-adapted
Concentration : 100 mg/l related to
related to
Contact time : 28 day(s)
Degradation : = 3 - 7 (±) % after 28 day(s)

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Result :
Deg. product :
Method :
Year :
GLP : no data
Test substance : other TS: Diethyl ether (no additional information)

Method : The volume of the test solution was 300 mL. The test was conducted at a test substance concentration of 100 mg/L and 25 deg C for 28 days.
Remark : In this test system, diethyl ether was not readily biodegradable.
Reliability : (2) valid with restrictions
Flag : Critical study for SIDS endpoint
30.11.2005 (22)

Type : aerobic
Inoculum : activated sludge, domestic, non-adapted
Concentration : 2 mg/l related to related to
Contact time : 28 day(s)
Degradation : = 5 (±) % after 28 day(s)
Result :
Deg. product :
Method :
Year :
GLP : no data
Test substance : other TS

Method : Directive 84/449/EEC, C.4 "Biotic degradation - modified AFNOR test NF T90/302". The inoculum used was activated sludge, domestic.
Remark : In an aerobic test, 2 mg/L of DME was 5% degraded after 28 days. Methane-utilizing microorganisms, abundantly present in nature, play a significant role in the removal of DME from aquatic ecosystems and soils.
Test substance : Dimethyl ether, purity not specified
Reliability : (4) not assignable
30.11.2005 (4)

Remark : For their studies concerning the microbic degradation of diethyl ether, Imai et al. (1986) used the thermophilic, obligate methane-oxidizing bacteria strain, "H-2", which was isolated from a gas field. This organism was taken from a continuously growing culture and employed without first being adapted. The measurable catabolite of diethyl ether degradation was acetic acid which was formed to 5,3 umol/h/mg protein. The authors attributed the fact that neither ethanol nor acetaldehyde was detected to a high dehydrogenase activity of the bacteria strain.
18.11.2003 (18)

3.6 BOD5, COD OR BOD5/COD RATIO

3.7 BIOACCUMULATION

3.8 ADDITIONAL REMARKS

4.1 ACUTE/PROLONGED TOXICITY TO FISH

Type : flow through
Species : Pimephales promelas (Fish, fresh water)
Exposure period : 96 hour(s)
Unit : mg/l
LC50 : = 2560 measured/nominal
EC50 : = 2260 measured/nominal
Method : other
Year : 1986
GLP : no data
Test substance : other TS

Method : [According to ASTM (1980), Standard practice for conducting acute toxicity tests with fish, macroinvertebrates and amphibians. American Society for Testing and Methods Committee E-35.]

Flow-through exposures were made with a continuous flow modified mini-diluter. One chemical stock solution was prepared and used for the entire test.

Gas-liquid chromatography (flame-ionisation detector) was used to analyze test substance concentrations in water samples from the exposure chambers. All test exposure chambers were sampled at approximately mid-depth at 0, 24, 48, 72 and 96 hours. All samples were analyzed immediately or adequately preserved for later analysis.

The fish were not fed 24 hours before or during the test. The tests were initiated by adding 20 fish per treatment and control groups. The number of dead fish was noted every 24 hours after the beginning of the test at which time they were also removed from the chambers. Observations of fish behavior and toxic signs were made at 2-8, 24, 48, 72 and 96 hours. Upon test termination, individual control fish were weighed (wet weight) and measured (standard length). Four surviving fish each from the control, the lowest concentration and the concentration nearest the LC50 were preserved for possible future histopathologic evaluation.

Result : The LC50 and EC50 values were calculated using the corrected averages of the analyzed tank concentrations and the Trimmed Spearman-Kärber Method. EC50's were based upon loss of equilibrium.

Analytical Results (in mg/l):

Nominal	Hours					
Conc. (mg/l)	0	24	48	72	96	Corrected Averages*
Control	<0.02	<0.02	<0.02	<0.02	<0.02	<0.02
1.96	0.487	0.382	0.428	0.350	0.429	0.40
3.02	0.577	0.624	0.681	0.750	0.769	0.66
4.65	0.861	1.14	1.26	1.27	1.27	1.12
7.15	1.58	1.58	2.02	2.05	1.91	1.76
11.0	2.72	2.57	3.33	3.18	3.06	2.87

*Corrected for analytical recoveries of spiked water samples.

Fish exposed to DEE lost schooling behavior and swam in a cork-screw/spiral pattern near the tank surface. They were underreactive to external stimuli, had increased respiration, were darkly colored and lost equilibrium prior to death. Differences in the measured and nominal tank values were due to volatilization of the chemical. Calculations were based on measured values.

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Cumulative Mortality (total number of animals in each group = 20):

Concentration (g/L)	Number of deaths Time (hours)			
	24	48	72	96
0	0	0	0	0
1.96	0	0	0	0
3.02	0	0	0	0
4.65	0	0	0	0
7.15	0	0	15	0
11.0	10	13	13	13

Number of animals with effects (total number of animals in each group = 20):

Concentration (g/L)	Time (hours)			
	24	48	72	96
0	0	0	0	0
1.96	0	0	0	0
3.02	0	0	0	0
4.65	0	0	0	0
7.15	0	0	0	0
11.0	20	20	20	20

Test condition

: Species: P. promelas,
 Age: 29 days,
 Weight: 0.069 +/- 0.0264 g,
 Length: 17.0 +/- 1.959 mm,
 Loading: 0.69 g/L
 Test medium: filtered Lake Superior water,
 Water quality parameters as measured during the test:
 Temperature = 24.8 degrees C;
 Dissolved oxygen = 7.1 mg/L;
 pH = 7.76;
 Total hardness = 45.1 mg/L CaCO₃; and
 Total alkalinity = 41.5 mg/L CaCO₃.

Test substance Reliability

: Diethyl Ether (CAS RN 60-29-7); Purity not specified.
 : (1) valid without restriction
 Similar to OECD 203

Flag

14.11.2005

: Critical study for SIDS endpoint

(47)

Type

: semistatic

Species

: Poecilia reticulata (Fish, fresh water)

Exposure period

: 14 day(s)

Unit

: mg/l

LC50

: = 2134

Limit test

:

Analytical monitoring

: no data

Method

: other: see text

Year

:

GLP

: no data

Test substance

: no data

Method

: The study measured the acute toxicity of the test substance to 2-3 month old guppies under static-renewal conditions for 14 days. The test substance was tested at several concentrations in a series with a 1.8-factor geometric progression. Stock solutions were prepared using a solvent (acetone or propanol-2) and diluted with standard water (hardness of 25

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	mg/l as CaCO ₃). Each test vessel contained approximately 1 L of test solution and eight guppies. Test solutions were renewed daily. Guppies were fed a commercial fish food 0.5 h before each renewal. The temperature and dissolved oxygen concentration during the test were maintained at 22±1°C and 5 mg/l, respectively. The guppies were considered to be dead when gill movements ceased and no reaction occurred when fish were touched with a glass bar.	
Source	:	Sodes Paris Huels AG Marl EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
Test condition	:	Testgefäesse abgedeckt, taeglicher Wasserwechsel, 22 Grad Celsius
Reliability 19.12.2003	:	(2) valid with restrictions (72)
Type	:	static
Species	:	Lepomis macrochirus (Fish, fresh water)
Exposure period	:	96 hour(s)
Unit	:	mg/l
LC0	:	>= 10000
Limit test	:	
Analytical monitoring	:	no
Method	:	
Year	:	
GLP	:	no data
Test substance	:	no data
Source	:	Sodes Paris Huels AG Marl EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
Test condition 19.12.2003	:	Open test system, 23 degree C (29)
Type	:	static
Species	:	Leuciscus idus (Fish, fresh water)
Exposure period	:	48 hour(s)
Unit	:	mg/l
LC0	:	= 2130
LC50	:	= 2840
LC100	:	= 3600
Limit test	:	
Analytical monitoring	:	no
Method	:	other: Bestimmung der Wirkung von Wasserinhaltsstoffen auf Fische, DIN38412 Teil 15
Year	:	
GLP	:	no data
Test substance	:	no data
Source 19.12.2003	:	Sodes Paris EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (68)
Type	:	semistatic
Species	:	Poecilia reticulata (Fish, fresh water)
Exposure period	:	96 hour(s)
Unit	:	mg/l
NOEC	:	> 4000 measured/nominal
Limit test	:	
Analytical monitoring	:	yes
Method	:	other
Year	:	1988
GLP	:	yes

4. Ecotoxicity

Id 60-29-7

Date 10.01.2006

- Test substance** : other TS
- Method** : NEN 6504; semistatic. With respect to rapid volatilization of DME, sealed flasks were used for the testing. Renewal of test solutions occurred after 48 hours. A total of 14 animals per concentration were tested in 2 replicates (7 animals/bottle x 2 bottle).
- Result** : All fish survived the dosages studied (nominal concentrations of 1900 and 3200 mg/L).

The table below presents additional information regarding water chemistry values obtained during the study.

pH	Not given at study start 7.3-7.5 at the end of test
DO	Saturated at study start 4.5-6.9 at the end of test
TOC	Not Given
Temperature	23°C
Water hardness	Not given

The measured concentrations obtained in the study can be found in the table below.

Nominal Concentration (ppm)	Measured Concentration (ppm)			
	0 hours	48 hours	Renewal 48 hours	96 hours
1900	665	675	1785	1845
1900	1150	1095	1640	1690
3200	4075	4140	4220	NM
3200	4180	4080	2085	NM

NM = No Measurement

- Test substance** : Dimethyl ether, purity 100%
- Reliability** : (2) valid with restrictions
- 14.11.2005

(3)

4.2 ACUTE TOXICITY TO AQUATIC INVERTEBRATES

- Type** : static
- Species** : Daphnia magna (Crustacea)
- Exposure period** : 24 hour(s)
- Unit** : mg/l
- EC50** : = 165
- Analytical monitoring** : no
- Method** : other: Daphnien-Kurzzeitest, DIN 38412 Teil 11, Bestimmung der Wirkung von Wasserinhaltsstoffen auf Kleinkrebse.
Gefaesse leicht abgedeckt
- Year** : 1982
- GLP** : no
- Test substance** : no data
- Remark** : Following is a summary of the test conditions:
Test type: Static

Test duration: 24 hours
Temperature: 20 degrees C
Light quality: Artificial light, OSRAM Neon light, Leuchtfarbe 25
Light intensity: $E_{0sy} = 2.5 \text{ W/m}^2$
Photoperiod: 9 hours
Feeding prior to test: Standardized dry algae
Feeding regime: None
Test chamber: 50 ml beakers filled with 20 ml liquid
Loading rate: 2 mL/daphnid
Test volume: Minimum 20 mL
Source: Standardized test strain IRCHA
Age of test organisms: 24 hours max.
Test concentrations: Not provided
Number of replicate test vessels per concentration in definitive test: 2 replicates per test and control concentration
Number of animals per replicate: 10
Aeration: None
Dilution water: Dilution water for culturing was tap water. Dilution water for testing was a chemically and physically defined standardized culture medium ("artificial fresh water").
Measured water chemistry parameters: Parameters determined at the end of the test:
pH (target: pH: 8.0 ± 0.2); dissolved oxygen (target: 2 mg/l); conductivity (not measured); temperature (constant 20 °C in incubator); visual observations at 24 hours; Dilution water hardness at 0 hours (not measured).
Measured endpoint: Immobility

Test concentrations were not provided; only the dilution ratios are indicated; 1st step: dilution ratio 1:2. 2nd step further dilution steps (1:1.4 or 1:1.1) in case no 3 grading between EC0 and EC100. The definitive test was based on a total of 20 daphnids per concentration tested (i.e., 10 daphnids per replicate, in each of 2 replicates) exposed to each test concentration, as well as a control (100% dilution water). Immobility and abnormal behavior (e.g., erratic swimming) were recorded at 24 hours. An EC50 (concentration causing immobility in 50% of the organisms) was estimated based on the 24-hour immobility data. The test was considered valid if immobility did not exceed 10% in the control. No reference that the control was also checked for this parameter.

Test conditions: Dilution water for testing was a chemically and physically defined standardized culture medium ("artificial fresh water" according to above DIN). Dilution water used for culturing was tap water. Test solutions were prepared as follows: The substances tested were poured in closed bottles containing the artificial fresh water on a magnetic stirrer until solution was optically transparent. Then the dilutions were made with this stock solution.

The 24-hour EC50 was calculated as follows: EC0 and EC100 values were determined. The percentage of immobile specimens was plotted against the concentration of the substances tested in mg/l (on Schleicher Schüll logarithmic paper No. 440 ½ A4; abscissa: the mg/l-concentration; ordinate: the percentage of immobilized daphnids). Ordinarily in the range of 16 - 84% immobilization the respective values should be found on a straight line. If values were on a straight line, then the EC50 value could be extrapolated and the 95% confidence range calculated. The authors also referenced the statistical method "Chi-Quadrat-Test". In cases where the slope was too steep and further testing of the 1:1.1 dilution ratio did not provide sufficient data points, then the geometric middle of the EC0 and EC100 was taken as the EC50 value.

Result
Source

: 24-hour EC50 = 165 mg/L
: Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

4. Ecotoxicity

Id 60-29-7

Date 10.01.2006

Reliability : (2) valid with restrictions
Flag : Critical study for SIDS endpoint
 19.12.2003

(14)

Type :
Species : Daphnia magna (Crustacea)
Exposure period : 48 hour(s)
Unit : mg/l
NOEC : > 4000 measured/nominal
Analytical monitoring : yes
Method : other
Year : 1988
GLP : yes
Test substance : other TS

Method : NEN 6501. With respect to rapid volatilization of DME, sealed flasks were used for the testing. A total of 12-14 animals per concentration were tested in 2 replicates (6-7 animals/bottle x 2 bottles).

Result : The table below presents additional information regarding water chemistry values obtained during the study.

pH	Not given at study start 7.3-8.1 at study end
DO	Saturated at study start >8 at study end
TOC	Not given
Temperature	20°C
Water Hardness	Not given

The measured concentrations obtained during the study can be found in the table below.

Nominal Concentration (ppm)	Measured Concentration (ppm)	
	0 hours	48 hours
1000	793	743
1000	263	312
3200	4135	4200
3200	4370	4385

All animals survived the dosages studied (nominal concentrations of 1000 and 3200 mg/L).

Test substance : Dimethyl ether, purity 100%.
Reliability : (1) valid without restriction
 14.11.2005

(3)

Type :
Species : Daphnia magna (Crustacea)
Exposure period : 48 hour(s)
Unit : mg/l
EC0 : = 1380
Analytical monitoring : no data
Method : other: see text
Year :
GLP : no data
Test substance : no data

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Id 60-29-7

Date 10.01.2006

Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
Test condition : unklar, ob offene oder geschlossene Testsysteme verwendet wurden
15.12.1993 (58)

4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

Species : other algae: Green Algae
Endpoint :
Exposure period : 96 hour(s)
Unit : mg/l
EC50 : = 410 calculated
Method : other: EPIWIN (v 3.11) ECOSAR Submodel (v 0.99g)
Year : 2003
GLP :
Test substance :

Remark : The EPIWIN model was run using the following measured physical chemical properties:
Water solubility (mg/L): 65000;
Vapor pressure (mm Hg): 442;
Log Kow (octanol-water): 0.82;
Boiling point (deg C): 34.50; and
Melting point (deg C): -116.20.
23.11.2005 (102)

4.4 TOXICITY TO MICROORGANISMS E.G. BACTERIA

Type : aquatic
Species : Photobacterium phosphoreum (Bacteria)
Exposure period : 15 minute(s)
Unit : mg/l
EC50 : = 5600
Analytical monitoring : no data
Method :
Year :
GLP : no data
Test substance : no data

Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
15.12.1993 (58)

4.5.1 CHRONIC TOXICITY TO FISH

4.5.2 CHRONIC TOXICITY TO AQUATIC INVERTEBRATES

4.6.1 TOXICITY TO SEDIMENT DWELLING ORGANISMS

4. Ecotoxicity

Id 60-29-7

Date 10.01.2006

4.6.2 TOXICITY TO TERRESTRIAL PLANTS

Species : other terrestrial plant: Mimosa pudica, Oxalis stricta, Marsilia macropus
Endpoint : other: inhibition opening / closing movements
Exposure period :
Unit : mg/l
Method :
Year :
GLP : no
Test substance : no data

Remark : EC100 = 330 - 510 mg/l
Effects were reversible after end of exposure within few hours.

Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

23.11.2005

(110)

4.6.3 TOXICITY TO SOIL DWELLING ORGANISMS

4.6.4 TOX. TO OTHER NON MAMM. TERR. SPECIES

4.7 BIOLOGICAL EFFECTS MONITORING

4.8 BIOTRANSFORMATION AND KINETICS

4.9 ADDITIONAL REMARKS

5. Toxicity

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Date 10.01.2006

5.0 TOXICOKINETICS, METABOLISM AND DISTRIBUTION

5.1.1 ACUTE ORAL TOXICITY

Type : LD50
Value :
Species : rat
Strain : Sprague-Dawley
Sex : male/female
Number of animals :
Vehicle : other: none
Doses :
Method :
Year : 1970
GLP : no
Test substance : no data

Remark : The test substance was administered undiluted to two groups of 6 nonfasted male Sprague-Dawley rats: Group 1 = young adults (80 - 160 g) and Group 2 = older adults (300 - 470 g). The test substance was also administered to groups of 6-12 nonfasted rats of both sexes at 14-days of age (16-50 g). The animals were observed for one week following dose administration. The LD50 and associated confidence limits were calculated both by the method of Litchfield and Wilcoxon (Litchfield, J.T. and F. Wilcoxon. 1949. A simplified method of evaluating dose-effect experiments. J. Pharmacol. Exp. Ther. 96:99-101) and by a probit analysis statistical program via an IBM-1800 calculator.

Result : Following are the LD50 values (95% confidence limits) for the individual age groups tested in this study:
14-day old rats: 1568 mg/kg (855 - 2352 mg/kg)
young adults: 1710 mg/kg (1425 - 1924 mg/kg)
older adults: 1211 mg/kg (1069 - 1354 mg/kg)

Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

Reliability : (2) valid with restrictions
Flag : Critical study for SIDS endpoint
23.11.2005 (70)

5.1.2 ACUTE INHALATION TOXICITY

Type : other: LT50
Value :
Species : rat
Strain : Sprague-Dawley
Sex :
Number of animals :
Vehicle :
Doses : 150,000 and 200,000 ppm (450 and 605 mg/L, respectively)
Exposure time :
Method :
Year : 1970
GLP : no
Test substance : no data

Remark : Number of animals: 10 adult females and 40 neonatal rats

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The median time to death (LT50 values) were calculated for adult and neonatal rats exposed to concentrations of ether vapor of 150,000 or 200,000 ppm. Animals were exposed in a 10 L vapor exposure chamber that was arranged as a closed circuit anesthesia apparatus that included a soda lime canister for absorption of carbon dioxide. Specific quantities of the test substance were vaporized in a 2 L polyethylene container in the circuit. The ether vapor was evenly distributed throughout the exposure chamber via the continuous flow of room air through the circuit by a pump. One adult non-pregnant female (275 - 325 g) and 4 neonatal rats (5 - 8 g) of either sex were exposed at a time, until a total of 10 adults and 40 neonatal rats were exposed. Animals were exposed to initial ether concentrations of 150,000 or 200,000 ppm (450 and 605 mg/L, respectively). The ether was not replenished throughout the exposure so the concentration of ether within the chamber gradually decreased as the exposure progressed. Ether concentrations were analyzed by use of a gas chromatograph and a flame ionization detector. Exposure chamber samples (100 µl) were obtained with a gas tight syringe through a rubber stoppered port in the chamber lid. Animals were observed, atmosphere samples were taken and blood ether concentrations were determined at 0.14 log time intervals. The blood ether concentration at the LT50 was determined from the blood ether-time plots constructed for neonates and adults.

Calculations of the median time to death (LT50) values for adults and neonatal rats were determined by the method of Litchfield, J.T. (A method of rapid graphic solution of time-percent effect curves. 1949. J. Pharmacol. Exp. Ther. 97: 399-408).

Result : The LT50 values (95% confidence limits) for adult and neonatal rats are outlined below:

	Number exposed	150,000 ppm LT50 (min.)	200,000 ppm LT50 (min.)
Adult	10	20 (18-24)	17 (14-20)
Neonate	40	135 (123-148)	86 (80-92)

Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

Reliability Flag : (2) valid with restrictions
: Critical study for SIDS endpoint

20.11.2003

(88)

Type : LC50
Value :
Species : mouse
Strain : other: C57BL/6
Sex : male/female
Number of animals :
Vehicle : other: none
Doses : 32,000 to 96,000 ppm (97 to 291 mg/L, respectively)
Exposure time : 90 minute(s)
Method :
Year : 1984
GLP : no data
Test substance : other TS: special grade diethyl ether from WAKO Pure Chemical Industries, Ltd., (Osaka)

Remark : In each trial, 5 or 6 mice were exposed for 120 min. to ether in a 14 L glass chamber connected to an anesthetic machine. Ether was vaporized and diluted with a fixed volume of air (6 L/min.) to result in several concentrations from 32,000 to 96,000 ppm (97 to 291 mg/L, respectively). Mortality was evaluated every 30 minutes during exposure and confirmed following exposure. The median lethal concentration (LC50) of ether was calculated using the data obtained after 90 minutes of exposure. This exposure time was chosen for the LC50 calculation because it was

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- considered to be the time point when the concentration of ether in the blood would be sufficiently in equilibrium with that of the vapor mixture. The LC50 values were determined using dose-mortality curves. Logarithm-probit transformation was employed for linearization of the dose-response curve. The LC50 values were estimated on the regression lines, and analysis of covariance was performed to examine the fitness of the lines and differences in susceptibility between the sexes.
- Number of animals: 10 to 15/sex/group
- Result** : Mortality of 4 week old mice after 90 min. of exposure:
- | Concentration (ppm) | Male | Female |
|---------------------|-------|--------|
| 32,000 | 0/10 | 0/10 |
| 46,000 | 0/10 | -- |
| 51,000 | 2/10 | -- |
| 55,000 | 2/10 | 0/10 |
| 60,000 | 4/15 | 3/10 |
| 66,000 | 9/15 | 5/10 |
| 73,000 | 10/10 | 8/10 |
| 80,000 | 10/10 | 10/10 |
| 96,000 | 5/5 | -- |
- LC50 (95% confidence limit):
Males: 60,000 ppm (54,500 - 66,100 ppm) or 182 mg/L (165 - 200 mg/L)
Females: 65,800 ppm (60,800 - 71,300 ppm) or 199 mg/L (184 - 216 mg/L)
- Source** : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
- Reliability** : (2) valid with restrictions
29.12.2003 (71)
- Type** : LC50
Value :
Species : mouse
Strain : other: C3H/He
Sex : male/female
Number of animals :
Vehicle :
Doses : 27,000 to 46,000 ppm (82 to 139 mg/L, respectively)
Exposure time : 90 minute(s)
Method :
Year : 1984
GLP : no data
Test substance : other TS: special grade diethyl ether from WAKO Pure Chemical Industries, Ltd., (Osaka)
- Remark** : Number of animals: 10 to 15/sex/group
In each trial, 5 or 6 mice were exposed for 120 min. to ether in a 14 L glass chamber connected to an anesthetic machine. Ether was vaporized and diluted with a fixed volume of air (6 L/min.) to result in several concentrations from 27,000 to 46,000 ppm (82 to 139 mg/L, respectively). Mortality was evaluated every 30 minutes during exposure and confirmed following exposure. The median lethal concentration (LC50) of ether was calculated using the data obtained after 90 minutes of exposure. This exposure time was chosen for the LC50 calculation because it was considered to be the time point when the concentration of ether in the blood would be sufficiently in equilibrium with that of the vapor mixture. The LC50 values were determined using dose-mortality curves. Logarithm-probit transformation was employed for linearization of the dose-response curve. The LC50 values were estimated on the regression lines, and analysis of covariance was performed to examine the fitness of the lines and differences in susceptibility between the sexes.
- Result** : Mortality of 4 week old mice after 90 min. of exposure:

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Concentration (ppm)	Male	Female
27,000	0/10	0/10
29,000	4/15	3/10
32,000	12/15	7/10
35,000	13/15	9/10
38,000	10/10	9/10
42,000	--	9/10
46,000	10/10	10/10

LC50 (95% confidence limit):

Males: 31,300 ppm (29,100 - 33,600 ppm) or 95 mg/L (88 - 102 mg/L)

Females: 32,400 ppm (28,900 - 36,200 ppm) or 98 mg/L (87 - 110 mg/L)

Source	: Sodes Paris Huels AG Marl EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
Reliability 29.12.2003	: (2) valid with restrictions (71)
Type	: LC50
Value	: = 164000 ppm
Species	: rat
Strain	: other
Sex	: male
Number of animals	: 50
Vehicle	: no data
Doses	: 84000, 121000, 152000, 169000 and 205000
Exposure time	: 4 hour(s)
Method	:
Year	: 1979
GLP	: no
Test substance	: other TS
Method	: Groups of 10 rats, 7 - 8 weeks old, were exposed to DME gas by whole-body method for single 4-hour periods. Exposure concentrations tested were 84,000, 121,000, 152,000, 169,000, and 205,000 ppm. Food and water were available ad libitum at all times other than the exposure. Atmospheres were generated by means of a single-stage regulator through a flow meter directly into the top of a 20-liter glass exposure chamber. Dilution air flowing through a flow meter joined the DME stream at the top of the chamber. The air/DME flow was maintained at 10 L/minute. Gas standards and samples were analyzed with a thermal conductivity detector on a gas chromatograph. Chamber atmosphere was sampled at approximately 30-minute intervals. During exposure, observations of clinical signs of toxicity were made. After exposure, the surviving rats were returned to their respective cages and were observed daily (weekends excluded) for 14 days. Body weights and clinical signs were recorded. Surviving rats were sacrificed after a 14-day recovery period. The LC50 of DME was calculated via probit analysis.
Remark	: Strain: ChR-CD
Result	: Mortality of 0/10, 3/10, 2/10, 7/10, and 7/10 occurred in the 84,000, 121,000, 152,000, 169,000, and 205,000 ppm groups, respectively. All but one death (205,000 ppm) occurred during the exposures. During exposure, the rats demonstrated ataxia (84,000 ppm and above), unresponsiveness to noise (121,000 ppm and above), anesthesia (84,000 ppm and above), paw waving (84,000 ppm), roving eyeballs (84,000 ppm), and coma (121,000 ppm and above). Ataxia was defined as uncoordinated. Anesthesia was considered unconsciousness with steady respirations (>50/min) and coma was considered unconsciousness with irregular, periodic or slow (<50/min) and shallow respirations. Post-exposure, survivors rapidly awoke and showed no clinical signs, other than transient

5. Toxicity

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weight loss for 1-2 days and sporadic lung noise.

Test substance : Dimethyl ether, purity 99.9%
Reliability : (1) valid without restriction
23.11.2005 (36)

Type : LC50
Value : = 130 mg/l
Species : mouse
Strain :
Sex :
Number of animals :
Vehicle :
Doses :
Exposure time : 3 hour(s)
Method : other: see reference
Year :
GLP : no
Test substance : no data

Remark : Number of animals exposed: 300.
Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
25.04.1994 (78)

Type : LCLo
Value : = 397 mg/l
Species : mouse
Strain :
Sex :
Number of animals :
Vehicle :
Doses :
Exposure time :
Method : other: see reference
Year :
GLP : no
Test substance : no data

Remark : Number of animals exposed: 4. Result: 99.2 mg/l slight
excitation, 198 mg/l deep anesthesia, 397 mg/l irregular
respiration and respiratory arrest.
Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
27.04.1994 (97)

Type : LCLo
Value : = 90 mg/l
Species : mouse
Strain :
Sex :
Number of animals :
Vehicle :
Doses :
Exposure time : 100 minute(s)
Method : other: see reference
Year :
GLP : no
Test substance : no data

Remark : Time of exposure: ca. 100 min; LC100: 128,34 mg/l.

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Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
25.04.1994 (69)

Type : LCLo
Value : = 329 mg/l
Species : rabbit
Strain :
Sex :
Number of animals :
Vehicle :
Doses :
Exposure time :
Method :
Year :
GLP : no
Test substance : no data

Remark : No data about time of exposure or number of animals provided.

Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
25.04.1994 (1)

Type : LCLo
Value :
Species : dog
Strain :
Sex :
Number of animals :
Vehicle :
Doses :
Exposure time : 71 minute(s)
Method : other: see reference
Year :
GLP : no
Test substance : no data

Remark : LCLo = 208 - 247 mg/L
Number of animals exposed: 20. Time of exposure: 20 - 120 min, average exposure time: 71 min.

Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
23.11.2005 (85)

Type : LCLo
Value : = 235 mg/l
Species : dog
Strain :
Sex :
Number of animals :
Vehicle :
Doses :
Exposure time :
Method : other: no data
Year :
GLP : no
Test substance : no data

5. Toxicity

Id 60-29-7

Date 10.01.2006

Remark : No data about time of exposure or number of animals provided.
Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
25.04.1994 (1)

5.1.3 ACUTE DERMAL TOXICITY

5.1.4 ACUTE TOXICITY, OTHER ROUTES

Type : LD50
Value : 2420 mg/kg bw
Species : mouse
Strain :
Sex :
Number of animals :
Vehicle :
Doses :
Route of admin. : i.p.
Exposure time :
Method : other: see reference
Year :
GLP : no data
Test substance : no data

Remark : Number of animals exposed: 12.
Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
25.04.1994 (35)

Type : LDLo
Value : = 2000 mg/kg bw
Species : guinea pig
Strain :
Sex :
Number of animals :
Vehicle :
Doses :
Route of admin. : i.p.
Exposure time :
Method : other: see reference
Year :
GLP : no
Test substance : no data

Remark : Number of animals exposed: 4.
Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
25.04.1994 (34)

Type : LDLo
Value : ca. 4290 mg/kg bw
Species : mouse
Strain :
Sex :

5. Toxicity

Id 60-29-7

Date 10.01.2006

Number of animals :
Vehicle :
Doses :
Route of admin. : s.c.
Exposure time :
Method :
Year :
GLP : no
Test substance : no data

Remark : Number of animals exposed not provided.
Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
25.04.1994 (1)

Type : LD50
Value : 996 mg/kg bw
Species : mouse
Strain :
Sex :
Number of animals :
Vehicle :
Doses :
Route of admin. : i.v.
Exposure time :
Method : other: see reference
Year :
GLP : no
Test substance : no data

Remark : Ether was dissolved in an emulsion (vehicle not provided)
and administered intravenously.
Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
14.11.2005 (33)

5.2.1 SKIN IRRITATION

Species : rabbit
Concentration :
Exposure :
Exposure time :
Number of animals :
Vehicle :
PDII :
Result :
Classification :
Method :
Year :
GLP : no
Test substance : no data

Remark : Dosage: 360 mg, open application, mild reaction. No further
data provided.
Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
27.04.1994 (106)

5. Toxicity

Id 60-29-7
Date 10.01.2006

Species : guinea pig
Concentration :
Exposure :
Exposure time :
Number of animals :
Vehicle :
PDII :
Result :
Classification :
Method :
Year :
GLP : no data
Test substance : no data

Remark : Dosage: 50 mg/24 h, severely irritating. No further data provided.

Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
25.04.1994 (59)

5.2.2 EYE IRRITATION

Species : rabbit
Concentration :
Dose :
Exposure time :
Comment :
Number of animals :
Vehicle :
Result :
Classification :
Method :
Year :
GLP : no
Test substance : no data

Remark : Dosage: 100 mg. Moderately irritating. No further data provided.

Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
25.04.1994 (43)

Species : rabbit
Concentration :
Dose :
Exposure time :
Comment :
Number of animals :
Vehicle :
Result :
Classification :
Method : other: see reference
Year :
GLP : no
Test substance : no data

Remark : Open, undiluted application of test substance. Result: slight reversible injury, grade 2 on a scale of 10.

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Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
25.04.1994 (92)

5.3 SENSITIZATION

Remark : A skin-sensitizing potential has not yet been detected.
Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
25.04.1994 (19)

5.4 REPEATED DOSE TOXICITY

Type :
Species : rat
Sex : male/female
Strain : Sprague-Dawley
Route of admin. : inhalation
Exposure period : 35 days
Frequency of treatm. : 24 hours/day, occasional interruptions of no more than 2 hours once per day
Post exposure period : none
Doses : 1000 ppm, 10,000 ppm (3.0 mg/L, 30 mg/L)
Control group : yes
NOAEL : = 30 mg/l
Method :
Year : 1975
GLP : no
Test substance : no data

Remark : Groups of 16 rats (equal number of male and female) were exposed to the anesthetic concentrations of diethyl ether (1000 or 10,000 ppm) continuously for 35 days. A control group of 72 rats were treated in a similar manner except they were not exposed to ether. Animals were acclimated to the chambers for five days prior to initiation of exposures. The rats were 150 to 275 g at study initiation. Air was circulated in the chambers by two routes: through a carbon dioxide (soda lime) absorber, and through an air conditioner. Measured oxygen concentrations were 21 to 24% and carbon dioxide levels, which were measured periodically by gas chromatography (GC), never exceeded 0.37%. The concentrations of the test atmospheres were measured automatically at four-hour intervals by GC. Any traces of test substances that may have been found in the control chamber were always less than 1/100 of the concentration in the experimental chamber. Body weights were measured on day 7, 14 and 35 of exposure. Blood was obtained from rats exposed to 10,000 ppm ether at the end of the exposure period. Hematocrit and erythrocyte, leukocyte and differential counts were measured.
After the 35-day exposure period, all animals were killed by CO2 inhalation. The heart, lungs, liver, kidney and spleen were weighed and retained. Pieces of skeletal muscle, jejunum, proximal femur and brain were also retained. All liver specimens were examined microscopically for the presence or absence of degenerative lesions, which included granular, vacuolar degeneration, zonal centrilobular lipodosis, focal lipodosis and focal necrosis.

Result : All rats survived 35 days of exposure. Ether treated animals revealed no significant deviation from the air-exposed controls in means of body weight,

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	liver-to-bodyweight ratio, blood morphology, histology. NOAEC = 30 mg/L (10,000 ppm).	
Source	: Sodes Paris Huels AG Marl EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)	
Reliability 23.11.2005	: (2) valid with restrictions	(94)
Type	:	
Species	: mouse	
Sex	: male/female	
Strain	: ICR	
Route of admin.	: inhalation	
Exposure period	: 35 days	
Frequency of treatm.	: 24 hour/day, occasional interruptions of no more than 2 hours, once per day	
Post exposure period	: none	
Doses	: 1000 ppm, 10000 ppm (3.0 mg/L, 30 mg/L)	
Control group	: yes	
NOAEL	: = 3 mg/l	
LOAEL	: = 30 mg/l	
Method	:	
Year	: 1975	
GLP	: no	
Test substance	: no data	
Remark	: Groups of 48 mice (equal number of male and female) were exposed to the anesthetic concentrations of diethyl ether (1000 or 10,000 ppm) continuously for 35 days. Two control groups of 32 animals each were also included. Animals were acclimated to the chambers for five days prior to initiation of exposures. The mice weighed 18 to 20 grams at study initiation. Air was circulated in the chambers by two routes: through a carbon dioxide (soda lime) absorber, and through an air conditioner. Measured oxygen concentrations were 21 to 24% and carbon dioxide levels, which were measured periodically by gas chromatography (GC), never exceeded 0.37%. The concentrations of the test atmospheres were measured automatically at four-hour intervals by GC. Any traces of test substances that may have been found in the control chamber were always less than 1/100 of the concentration in the experimental chamber. Body weights were measured on day 7, 14 and 35 of exposure. After the 35-day exposure period, all surviving animals were killed by CO2 inhalation. The heart, lungs, liver, kidney and spleen were weighed and retained. Pieces of skeletal muscle, jejunum, proximal femur and brain were also retained. All liver specimens were examined microscopically for the presence or absence of degenerative lesions, which included granular, vacuolar degeneration, zonal centrilobular lipidosis, focal lipidosis and focal necrosis.	
Result	: 10,000 ppm: By exposure day 20, 25% of the mice in the 10,000 ppm exposure group died; therefore, surviving animals in this group were killed on day 20. Animals showed statistically significant increases in liver weight and liver-to-body weight ratio, no other observed parameter was affected. 1,000 ppm: Treated animals revealed no significant deviation from the air-exposed controls in means of body weight, blood morphology or histology. Liver weight and liver-to-bodyweight ratios were significantly increased in the male mice compared to controls.	
Source	: NOAEC = 3.0 mg/l (1,000 ppm); LOAEC = 30 mg/l (10,000 ppm). Sodes Paris Huels AG Marl EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)	
Reliability	: (2) valid with restrictions	

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(94)

Type :
Species : guinea pig
Sex : male/female
Strain : Hartley
Route of admin. : inhalation
Exposure period : 35 days
Frequency of treatm. : 24 hour/day, occasional interruptions of no more than 2 hours, once per day
Post exposure period : none
Doses : 1000 ppm, 10000 ppm (3.0 mg/L, 30 mg/L)
Control group : yes
NOAEL : = 3 mg/l
LOAEL : = 30 mg/l
Method :
Year : 1975
GLP : no
Test substance : no data

Remark : Groups of 16 guinea pigs (equal number of male and female) were exposed to the anesthetic concentrations of diethyl ether (1000 or 10,000 ppm) continuously for 35 days. Two control groups of 8 animals each were also included. Animals were acclimated to the chambers for five days prior to initiation of exposures. The guinea pigs weighed 250 to 350 grams at study initiation. Air was circulated in the chambers by two routes: through a carbon dioxide (soda lime) absorber, and through an air conditioner. Measured oxygen concentrations were 21 to 24% and carbon dioxide levels, which were measured periodically by gas chromatography (GC), never exceeded 0.37%. The concentrations of the test atmospheres were measured automatically at four-hour intervals by GC. Any traces of test substances that may have been found in the control chamber were always less than 1/100 of the concentration in the experimental chamber. Body weights were measured on day 7, 14 and 35 of exposure. After the 35-day exposure period, all surviving animals were killed by CO₂ inhalation. The heart, lungs, liver, kidney and spleen were weighed and retained. Pieces of skeletal muscle, jejunum, proximal femur and brain were also retained. All liver specimens were examined microscopically for the presence or absence of degenerative lesions, which included granular, vacuolar degeneration, zonal centrilobular lipidosis, focal lipidosis and focal necrosis.

Result : 10,000 ppm: By exposure day 20, 25% of the guinea pigs in the 10,000 ppm exposure group died; therefore, surviving animals in this group were killed on day 20. Animals showed reduced body weight gain; no other observed parameter was affected.
1,000 ppm: treated animals revealed no significant deviation from the air-exposed controls in means of body weight, liver-to-bodyweight ratio, blood morphology, or histology.

Source : NOAEC = 3.0 mg/l (1,000 ppm); LOAEC = 30 mg/l (10,000 ppm).
Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
Reliability : (2) valid with restrictions

23.11.2005

(94)

Type :
Species : rat
Sex : male/female
Strain : other: albino, not specified
Route of admin. : gavage
Exposure period : 90 days

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Frequency of treatm.	: no data
Post exposure period	: no data
Doses	: 500, 2000, 3500 mg/kg bw d
Control group	: yes
NOAEL	: = 500 mg/kg bw
LOAEL	: = 2000 mg/kg bw
Method	: other: see reference
Year	:
GLP	: no data
Test substance	: no data
Remark	: 30 animals/dose/sex; four dose levels: 0, 500, 2000, and 3500 mg/kg bw d
Result	: At 3500 mg/kg bw d, 15/60 rats died, there were observed inhibition in body weight gain and decreased food consumption, decreases in hemoglobin and hematocrit values, and a slight increase in red blood cell count. SGPT (= SALT, Serum alanine amino transferase) and serum cholesterol levels were significantly increased. At 2000 mg/kg bw d, 4/60 rats died, and inhibitions in body weight gain, transient increases in serum cholesterol, retinal atrophy, elevated relative hepatic weights, and gross necropsy aberrations were observed. At 500 mg/kg bw d, one rat had retinal atrophy, but no other effects of histopathologic lesions were observed. Thus 500 mg/kg bw d might be considered as a NOEL.
Source	: Sodes Paris Huels AG Marl EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
Reliability	: (4) not assignable
20.11.2003	(5)
Type	: Chronic
Species	: rat
Sex	: male/female
Strain	: other
Route of admin.	: inhalation
Exposure period	: 2 years
Frequency of treatm.	: 6 hours/day, 5 days/week (excluding holidays)
Post exposure period	:
Doses	: 0, 2000, 10000, and 25000 ppm
Control group	: yes, concurrent vehicle
NOAEL	: = 2000 ppm
Method	:
Year	: 1986
GLP	: yes
Test substance	: other TS
Method	: Groups of 100 male and 100 female rats were exposed to 0, 2000, 10,000, or 25,000 ppm DME for up to 2 years. Food and water were available to the rats ad libitum except during exposures. The age of rats was not specified. Rats were exposed whole-body to the vapor. During exposures, chamber temperature and relative humidity were maintained at approximately 23±2°C and 50±10%, respectively. DME vapors were generated by warming the compressed-gas cylinders containing liquefied DME in a 21-27 °C water bath. The vapors were metered into the intake manifold at the top of the exposure chamber. Filtered, conditioned air also entered the top of the chamber, swept the test material into respective exposure chambers, and was exhausted out the bottom of the chambers. Chamber concentrations of DME were regulated by controlling the flow rate of DME vapors into the chamber. Filtered air, alone, was metered in a similar manner into the control chamber. Total flow of air (control group) or

air plus DME was maintained at approximately 800 L/minute. Chamber atmospheres were quantitatively analyzed for DME by gas chromatography.

All rats were weighed and individually handled and carefully examined for abnormal behavior and appearance once weekly during the first 3 months of the study and twice monthly for the remainder of the study. Cage-site examinations to detect moribund or dead rats and abnormal behavior and appearance were conducted at least twice daily throughout the study. Approximately 3, 6, 9, 12, and 18 months after the study's initiation, hematological, clinical chemical, and urine analytical evaluations were conducted on 10 male and 10 female rats randomly selected from each exposure group. Fourteen hematological and 10 clinical chemistry parameters were measured or calculated. On the day prior to each bleeding time, an overnight urine specimen was collected and 9 urine chemistry parameters were measured or calculated. Gross and histopathological evaluations were conducted on 10 rats/sex/exposure group after 6, 12, and 18 months of exposure and on all rats alive after 2 years of exposure. Approximately 50 organs and/or tissues were saved for microscopic examinations. Organ weights were recorded on 10.

**Remark
Result**

- : Strain: Crl: CD(SD)BR
- : The overall mean weekly chamber concentrations of DME vapors were 2100 ± 200 , $10,200 \pm 900$, and $24,700 \pm 1900$ ppm for the 2000, 10,000, and 25,000 ppm groups, respectively.

Body weights were greater and survival rates were less than the control group for male rats in the 10,000 and 25,000 ppm DME groups. No clear association could be made between body weight increases and decreased survival even though these changes were concurrent observations in the same exposure groups. No histological lesion was found that could explain the decrease in survival rate. Body weights and survival rate of the female rats were statistically the same as the female rats in the control group.

Increased incidences of stained wet/perineal area were observed in male rats in the groups exposed to DME vapors. Since increases were observed in male rats in all exposure groups and since these increases were not exposure-related, the significance of this finding was not clear. Increased incidences of torn ears were observed in the male and female rats in the 10,000 and 25,000 ppm groups. Ear punching was used to identify the animals in the study. The 25,000 ppm rats had double punching of one ear, and the 10,000 ppm rats had single punching in both ears and this may have led to an increased incidence of torn ears in these groups.

Compound-related hematologic or clinical chemistry effects were not observed for male rats exposed to DME vapors for 2 years. A compound-related hemolytic effect was observed in male rats in the high-exposure group at 6 months on test. This effect was characterized by a decrease in erythrocyte count, increases in spleen weight, histological evidence of splenic congestion, along with normal bone marrow histology. A decrease in erythrocyte count was also observed in female rats at the high-exposure group at 3 months that was considered compound-related. These changes were interpreted to be transient effects that were not representative of the long-term effects of DME.

The incidence of clinically observable masses in female rats was higher in the 2000, 10,000, and 25,000 ppm groups. The masses were primarily ventral (axillary, inguinal, and perineal). An increase in the incidence of mammary tumors (benign or malignant) was observed in female rats in the 25,000 ppm DME group. These incidences of ventral masses and mammary tumors were considered not to be compound related because the incidences of masses and tumors in the control group were

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uncharacteristically low in comparison with the control groups incidence in studies previously conducted at Haskell Laboratory.

Exposure Group: (ppm)	0	2000	10,000	25,000
# Rats/Group	78	79	77	75
# Rats historically examined:	75	77	74	70
# Rats w/>= 1 benign mammary tumor:	16	30*	24	29*
# Rats w/>= 1 malig- nant mammary tumor:	14	16	16	20
# Rats w/>= 1 benign or malignant mammary: tumor:	27	34	35	37*
% Rats w/>= 1 benign or malignant mammary tumor:@	36.0	44.2	47.3	52.9

* = Statistically different from the control group ($p < 0.05$) by the Fisher's Exact test.

@ = Percentages were not analyzed statistically.

The increased incidences of benign tumors in the 2000 and 25,000 ppm groups were considered not to be biologically significant because of the lack of correlation with exposure concentration and because of inherent difficulties in correctly diagnosing tumors as benign or malignant. Thus, instead of considering specific tumor type, the total number of rats with at least one benign or malignant tumor was used for comparison of the incidence of mammary tumors in female rats. This comparison revealed a statistically significant ($p = 0.03$) increase in the incidence in the 25,000 ppm group when using the Fisher's Exact test. Whereas this incidence was significantly greater, the biological significance of this difference was questioned after comparison with the historical control group data. In five long-term inhalation studies conducted at Haskell Laboratory between 1980 and 1985, the overall incidence of control group female rats with at least one benign or malignant mammary tumor was 53%. The number of rats with benign or malignant mammary tumors and the percentage in parentheses for the five studies was: 54/87 (62.1%), 62/115 (53.9%), 38/86 (44.2%), 39/77 (57.1%), and 33/71 (46.5%). Thus, the incidence of mammary tumors for female rats exposed to DME vapors was similar to the mammary tumor incidence reported in the long-term inhalation studies conducted at Haskell Laboratory. The statistically significant increase in mammary tumors observed was considered not to be compound related. Rather, the control group incidence was uncharacteristically low in comparison with historical control group incidence.

No DME-related histological lesion was consistently observed throughout the study.

The no-observable-effect-concentration (NOEC) was 2000 ppm DME based on an increase in body weight and a decrease in survival in male rats exposed to 10,000 or 25,000 ppm DME vapors and on hemolytic effects noted in male rats exposed to 25,000 ppm DME vapors for 6 months. No neoplastic lesions were observed that could be attributable to

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Test substance : DME exposure. DME was not carcinogenic.
Reliability : Dimethyl ether, purity 99.98%
23.11.2005 : (1) valid without restriction (38)

Type :
Species : other: rat/guinea pig/rabbit
Sex : male/female
Strain : other: Wistar/-/
Route of admin. : inhalation
Exposure period : 7 weeks
Frequency of treatm. : 5 days/week, 7 hours/day
Post exposure period : none
Doses : 2000 ppm (6.2 mg/l)
Control group : yes
NOAEL : = 6.2 - mg/l
Method : other: see reference
Year :
GLP : no
Test substance : no data

Remark : approximate group sizes: 20 rats, 10 guinea pigs, 4 rabbits, equally divided as to sex.

Result : Ether treated animals revealed no deviation from the air-exposed controls in means of general toxicity, body weight, organ-to-bodyweight ratio, hematological parameters, SGOT and SGPT (= SAST and SALT, serum aspartate amino transferase and serum alanine transferase), histology.

Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

Reliability : (4) not assignable
14.11.2005 (24)

5.5 GENETIC TOXICITY 'IN VITRO'

Type : Ames test
System of testing : Salmonella typhimurium TA 1535, TA 100, TA 1538, TA 98, TA 1537
Test concentration :
Cytotoxic concentr. :
Metabolic activation : with and without
Result : negative
Method : other: Ames (1975); Maron and Ames (1983)
Year : 1984
GLP : no data
Test substance : other TS: ethyl ether

Remark : Ethyl ether was tested with each strain in duplicate or triplicate using the plate-incorporation method both with and without S9 activation. The test concentrations varied starting from the solubility or toxicity limit of the test substance. The S9 mix contained 10% liver S9 fractions from Aroclor-treated Sprague-Dawley rats, whose protein concentration had been adjusted to 30 mg/ml. The criteria for a positive response included rate of increase of induced versus spontaneous revertants, dose dependency, and reproducibility of results.

Result : The spontaneous reversions rates of the tester strains were within the expected ranges throughout the experiment.
Ethyl ether did not increase the number of revertants in Salmonella typhimurium.

Source : Sodes Paris

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Reliability	:	Huels AG Marl	
Flag	:	EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)	
14.11.2003	:	(2) valid with restrictions	
	:	Critical study for SIDS endpoint	(30)
Type	:	DNA damage and repair assay	
System of testing	:	E. coli WP2, WP67, CM871	
Test concentration	:		
Cycotoxic concentr.	:		
Metabolic activation	:	with and without	
Result	:	negative	
Method	:	other: Liquid micromethod procedure	
Year	:		
GLP	:	no data	
Test substance	:	other TS: ethyl ether	
Remark	:	<p>Disposable, sterile Microtiter plates containing 8 rows of 12 350 µl wells were utilized for this test. Fifty microliters of the test substance were distributed in each of the first wells of six 8-well rows. The test concentrations varied beginning with the solubility or toxicity limit of the test substance. With this as the starting concentration, the test substance was further diluted in nutrient broth for a total of eight, 2-fold dilutions (50 µl/well, 6 wells/dilution). Of these six 8-well rows, three were filled with 50 µl 0.2 M phosphate-buffered saline (PBS) and 3 with 50 µl S9 mix. The S9 mix contained 10% liver S9 fractions from Aroclor-treated Sprague-Dawley rats, whose protein concentration had been adjusted to 30 mg/ml. Finally, each row of wells was filled with 100 µl of one of the three bacterial strains. The plates were sealed with self-adhesive acetate tape and then place on a multiple microshaker apparatus for 5 minutes of mixing. Bacterial growth in each well was visually evaluated after 16 hours at 37 degrees C, by observing the increase in turbidity of the medium and/or formation of a pellet of settled cells on the bottom of the wells. Five separate experiments were performed in order to determine the reproducibility of results. The test was considered positive if the ratio between the minimal inhibitory concentrations (MICs) in repair-proficient (WP2) and repair-deficient (Wp67, CM871) tester strains were greater than 2.</p>	
Result	:	Ethyl ether did not cause genotoxicity in E. coli strains deficient in tryptophan synthesis. The MIC of all tester strains both with and without S9 mix were identical (i.e >40,000 µg).	
Source	:	Sodes Paris	
	:	Huels AG Marl	
	:	EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)	
Reliability	:	(2) valid with restrictions	(30)
14.11.2003	:		
Type	:	Ames test	
System of testing	:	Salmonella typhimurium TA 100, TA 98	
Test concentration	:	1, 5, or 10 %	
Cycotoxic concentr.	:		
Metabolic activation	:	with and without	
Result	:	negative	
Method	:	other: see reference	
Year	:		
GLP	:	no	
Test substance	:	no data	
Source	:	Sodes Paris	
	:	Huels AG Marl	
	:	EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)	
25.04.1994	:		(111)

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Type : Sister chromatid exchange assay
System of testing : Chinese hamster ovary (CHO) cells
Test concentration : 1,97%
Cycotoxic concentr. :
Metabolic activation : with and without
Result : negative
Method : other: see reference
Year :
GLP : no
Test substance : no data

Source : Sodes Paris
 Huels AG Marl
 EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
 25.04.1994 (115)

Type : Sister chromatid exchange assay
System of testing : Chinese hamster ovary (CHO) cells
Test concentration : no data
Cycotoxic concentr. :
Metabolic activation : no data
Result : negative
Method : other: no data
Year :
GLP : no data
Test substance : no data

Remark : Ethyl ether had no effect on the number of sister chromatid exchanges in cultured Chinese hamster ovary cells.
Source : Sodes Paris
 Huels AG Marl
 EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
 26.04.1994 (2)

Type : Bacterial reverse mutation assay
System of testing : Salmonella typhimurium strains TA97a, TA98, TA100 and TA1535 and Escherichia coli strain WP2uvrA (pKM101)
Test concentration : Trials 1 and 2: 0, 20, 30, 40, 50, 75%
 Trial 3: 0, 45, 55, 65%
Cycotoxic concentr. :
Metabolic activation : with and without
Result : negative
Method : other
Year : 2000
GLP : yes
Test substance : other TS

Method : This study followed the following test guidelines:
 U.S. EPA Health Effects Test Guidelines OPPTS 799.9510 (1989)
 OECD Guidelines for Testing of Chemicals Section 4: Health Effects, No. 471 (Adopted 1997)
 Commission Directive 92/69/EEC, EEC Method B.12
 The study consisted of 2 independent trials with and without a metabolic activation system. A third trial, utilizing S. typhimurium TA98 with S9 was used to confirm the results. Three replicates were plated for each tester strain, test concentration, and condition. Positive and negative controls were included in all assays. The reaction mixture (S-9 mix) contained glucose 6-phosphate, NADP, NaH2PO4, KCL, MgCL2, distilled water, and

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S-9. Treatments with activation were conducted by adding 0.5 mL of S-9 mix, and 0.1 mL of an overnight culture to 2 mL of top agar. These components were briefly mixed and poured onto a minimal glucose agar plate. Treatments in the absence of the metabolic activation system were identical to those with activation with the exception that 0.5 mL of sterile buffer was used as a replacement for the S-9.

Plates were exposed to dilutions of the test gas in 6-L gas chambers. The test substance and filtered air flows were regulated using individual rotameters, and mixed prior to entry into the chambers. Chambers were placed into an incubator at 37°C for approximately 48 hours. Gas chromatographic analysis was used to confirm the concentration of test atmospheres.

Bacterial background lawns were evaluated for evidence of test substance toxicity and precipitation. Revertant colonies for a given tester strain and condition were counted by an automated colony counter.

Positive control substances tested in this study included 2-nitrofluorene, N-ethyl-N-nitro-N-nitroguanidine, sodium azide, ICR 191 acridine mutagen, 9,10-dimethyl-1,2-benzanthracene, and 2-aminoanthracene.

Filtered house-line air was the test substance diluent and negative control.

A test substance was classified as positive if the mean number of revertants in any strain (except *S. typhimurium* TA1535) at any concentration was at least 2 times greater than the mean number of revertants of the concurrent negative control, and there was a concentration-related increase in the mean number of revertants per plate in that same strain. For *S. typhimurium* TA1535, there must be no test substance concentration with a mean number of revertants that is at least 3 times greater than the mean number of revertants of its concurrent negative control and a concentration-related increase in the mean number of revertants per plate. A test substance was classified as negative if all positive classification criteria for all strains are not met. Results not meeting criteria for either positive or negative classification were evaluated using scientific judgement and experience and may have been reported as equivocal.

Remark

: In trial 1, there was an apparent chamber leakage in one chamber at the high dose without S-9. The other chamber concentrations decreased approximately 50% from the mean at 0-hr and 48-hr. Test substance-related toxicity, as evidenced by a concentration dependent reduction in mean revertant colonies per plate, was observed in all tester strains except *S. typhimurium* strains TA100 and TA1535 without S-9. No evidence of mutagenicity was observed.

In trial 2, test substance-related toxicity was observed in all tester strains in the presence and absence of the metabolic activation system. The chamber concentrations decreased approximately 36% after 48 hours. No mutagenicity was observed with the exception of an equivocal response in *S. typhimurium* strain TA98 in the presence of S-9. At the 50% target concentration, a doubling of mean revertant plate count was observed compared to the mean of the concurrent negative control. There was no concentration-related increase in the tester strain, therefore, the data were considered inconclusive, and a third trial was initiated.

In trial 3, a mean decrease in chamber concentration of approximately 22% was observed with no apparent chamber leakage. Since this trial was negative with evidence of toxicity, the conclusion from trials 2 and 3 is that no evidence of mutagenicity was affirmed.

All acceptability criteria were met in this test. All tester strains exhibited

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appropriate phenotypic characteristics. No test substance-related precipitate was observed. The mean number of revertants in the negative control for each strain was within the prescribed acceptable historical control range. Mean positive control values for the tester strains exhibited 3-fold increase over the means of the respective negative controls for each test strain. Differences between targeted and actual doses in both analyses were acceptable for the purposes of this assay and in no way impacted the integrity or validity of this study.

Result : negative
Test substance : Dimethyl ether, 99.8%
Reliability : (1) valid without restriction
23.11.2005

(39)

Type : Chromosomal aberration test
System of testing : Human lymphocytes
Test concentration : Test 1 (3-hour exposure with and without S-9): 0, 35, 50, 70%; Test 2 (3-hour exposure with S-9): 0, 35, 50, 70%; Test 2 (19-hour exposure without S-9): 0, 20, 35, 50%

Cytotoxic concentr. :
Metabolic activation : with and without
Result : negative
Method : other
Year : 2000
GLP : yes
Test substance : other TS

Method : This study followed the following test guidelines:

U.S. EPA Health Effects Test Guidelines OPPTS 870.5375 (1998)

OECD Guidelines for Testing of Chemicals Section 4:
Health Effects, No. 473 (Adopted 1997)

Human lymphocytes, in whole blood culture, were stimulated to divide by addition of phytohaemagglutinin, and duplicate cultures were exposed to the test substance. Treatment atmospheres of the test substance were prepared in sterile glass bottles with septum caps. Negative and positive control cultures were also prepared. Mitomycin C and cyclophosphamide were used as positive control substances. Air was used as the negative control substance.

The test substance was sampled from the cylinder into a gas-sampling bag. Air was withdrawn from each pre-warmed (37°C) bottle and then an appropriate volume of test substance gas was introduced from the sampling bag, inserted through the septum cap, and the atmosphere was equilibrated at 37°C. After injection of the lymphocyte culture, air was allowed to enter each bottle through a hollow needle to produce the required concentration at atmospheric pressure. After approximately 48 hours, the cultures in duplicate were injected into the sterile glass bottles. The culture bottles were incubated on their sides at 37°C in a roller apparatus which rotated the bottles once every 8 minutes.

Test 1 included a 3-hour treatment with and without S-9 mix and 16 hours of recovery. Test 2 included a 3-hour treatment with S-9 mix and 16 hours of recovery. Test 2 also included a 19-hour continuous treatment without S-9.

Two hours before the end of the incubation period, cell division was arrested using Colcemid®, the cells harvested and slides prepared, so that metaphase cells could be examined for numerical (polyploidy) and structural chromosomal damage.

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In order to assess the toxicity to cultured lymphocytes, the mitotic index was calculated for all cultures treated with the test substance and the negative control. The highest dose level scored for chromosomal damage was, whenever possible, selected as the dose level causing a relative depression in mitotic index of at least 50%.

The test substance was considered to cause a positive response if the following conditions were met:

Statistically significant increases ($p < 0.01$) in the frequency of metaphases with aberrant chromosomes (excluding gaps) were observed at one or more test concentration.

The increases exceeded the negative control range of this laboratory, taken at the 99% confidence limit.

The increases were reproducible between replicate cultures.

The increases were not associated with large changes in osmolality of the treatment medium or extreme toxicity.

Evidence of a dose-relationship was considered to support the conclusion.

A negative response was claimed if no statistically significant increases in the number of aberrant cells above concurrent control frequencies were observed, at any dose level.

Remark

- : A relative depression in mitotic index of at least 50% was observed only at the top two dose levels after the 19-hour exposure in the absence of S-9 mix. The relative mitotic index was 44% and 18% at the test substance dose levels of 50 % and 70 %, respectively. The 50% dose level was selected as the top dose for chromosomal aberration analyses.

In both the absence and presence of S-9 mix, the test substance caused no statistically significant increase in the proportion of metaphase figures containing chromosomal aberrations, at any dose level, when compared with the negative control, in either test.

No increases in the proportion of polyploid cells were seen in the first test with 3-hour exposure in the absence of S-9 mix. However, in the presence of S-9 mix, a small statistically significant increase in the proportion of polyploid cells was seen at the highest level. In the second test both in the absence (19-hour exposure) and presence (3-hour exposure) of S-9 mix, the test substance caused small statistically significant increases in the proportion of polyploid metaphases at the highest level analyzed. This may indicate that the test substance has the potential to inhibit mitotic processes and to induce numerical chromosome aberrations.

All positive control compounds caused large, statistically significant increases in the proportion of aberrant cells, demonstrating the sensitivity of the test system and the efficacy of the S-9 mix.

Result

Test substance

Reliability

14.11.2005

- : negative
: Dimethyl ether, purity 100%
: (1) valid without restriction

(40)

5.6 GENETIC TOXICITY 'IN VIVO'

5.7 CARCINOGENICITY

5.8.1 TOXICITY TO FERTILITY

Type :
 Species : mouse
 Sex : male
 Strain : other: (C75BlxC3H)F1
 Route of admin. : inhalation
 Exposure period : 5 days
 Frequency of treatm. : 4 hours/day
 Premating exposure period :
 Male :
 Female :
 Duration of test :
 No. of generation :
 studies :
 Doses : 3,200 and 16,000 ppm (9.7 and 48 mg/L)
 Control group : yes
 Method : other: see remark
 Year :
 GLP : no data
 Test substance : no data

Remark : Five male mice per group, 11 weeks of age, were exposed to diethyl ether vapors four hours/day, for 5 consecutive days. Exposure chambers were constructed from 5-Liter glass desiccators with fenestrated porcelain floors. The test substance was delivered in air from calibrated vaporizers and entered the chamber below the floor and exhausted near the top of the chamber. The total flow of fresh gas to the chamber during exposure was 2.5 L/min. Three separate control groups of 5 mice each (15 total) were exposed to air under identical conditions as the test group. Each 4-hour exposure period was followed by a recovery period of one hour in air before animals were returned to their cages. The concentration of the vaporized test substance in the atmosphere and the CO₂ concentration of the exhausted chamber air were monitored periodically by gas chromatography. Chamber temperatures were also measured. The mice were killed 28 days after the first exposure day. Both cauda epididymides were removed, minced with scissors into 2 ml physiologic saline, pipetted and filtered through stainless gauze. The filtered suspension was stained overnight and duplicate slides were made for each animal. One thousand spermatozoa were examined on each slide for morphological abnormalities. All slides were read without knowledge of dose level. The number of morphologically abnormal cells was reported in percentages.

Result : The measured concentrations were within 5% of the target concentrations. The CO₂ concentration of the exhaust gas was maintained below 0.3% throughout all exposures. One mouse in the 3,200 ppm group did not survive the exposures. All surviving mice were evaluated for morphologically abnormal spermatozoa. No increase in the number of abnormal epididymal spermatozoa were found when compared to the control group as the following table indicates:

Group	Concentration (ppm)	Percent abnormal spermatozoa (+/- SEM)
Control	0	1.42 (+/- 0.08)
DEE	16,000	1.24 (+/- 0.11)
DEE	3,200	1.70 (+/- 0.23)

Source : Sodes Paris
 Huels AG Marl
 EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

5. Toxicity

Id 60-29-7

Date 10.01.2006

Reliability : (2) valid with restrictions
20.11.2003

(74)

5.8.2 DEVELOPMENTAL TOXICITY/TERATOGENICITY

Species : rat
Sex : female
Strain : Sprague-Dawley
Route of admin. : inhalation
Exposure period : 1 hour
Frequency of treatm. : days 9, 10, 11, or days 13, 14, 15 of gestation
Duration of test :
Doses : 73,000 ppm (220 mg/L)
Control group : no data specified
Method : other: see remark
Year :
GLP : no
Test substance : no data

Remark : Pregnant Sprague-Dawley rats were anesthetized with 7.3 vol% ether during early or late organogenesis. Animals were anesthetized for 1 hour in a 5.0 liter closed circuit anesthesia apparatus. Litters were delivered by cesarean section on day 19 of gestation, weighed, examined and measured.

Result : Ether anesthesia of pregnant rats caused early and late fetal resorptions and skeletal anomalies but did not alter the incidence of soft tissue anomalies. Thus ether did not show to be highly teratogenic, hypoxia might contribute to the embryotoxicity of ether.

Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

Reliability : (2) valid with restrictions
14.11.2005

(86)

Species : rat
Sex : female
Strain : Sprague-Dawley
Route of admin. : inhalation
Exposure period : 60-360 min
Frequency of treatm. : no data
Duration of test : no data
Doses : 73,000 ppm
Control group : no data specified
Method : other: see remark
Year :
GLP : no
Test substance : no data

Remark : In a preliminary study, rats had been exposed to 7.3 vol% diethyl ether for various lengths of time (60-360 min). Twenty-four hours later, the number of dead per group was counted. Fifty percent of the rats died after 150 min anesthesia.
Pregnant Sprague-Dawley rats (number of animals not provided) were then anesthetized for one hour in a 5 liter closed circuit vapor exposure chamber with 7.3 vol% diethyl ether during early or late embryogenesis. Fetuses were delivered by cesarean section one day before normal

5. Toxicity

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Result	: parturition. : Ether anesthesia (at 7.3 vol% for 60 min) of pregnant rats did not increase the incidence of resorptions, of soft tissue or skeletal anomalies. Anesthesia during early or late organogenesis did significantly decrease fetal bodyweight and length of long bones. Histologic examination of fetal brain, heart, kidney, liver, and skeletal muscle revealed no changes. Rats are more resistant than mice to the embryotoxic effects of ether anesthesia, however, ether is not highly embryotoxic to either mice or rats.
Source	: Sodes Paris : Huels AG Marl : EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
Reliability 18.11.2003	: (2) valid with restrictions (87)
Species	: rat
Sex	: female
Strain	: other
Route of admin.	: inhalation
Exposure period	: Days 6-15 of gestation; Cesarean section Gestation Day 21
Frequency of treatm.	: 6 hours/day
Duration of test	:
Doses	: 0, 1250, 5000, and 20000 ppm
Control group	: yes, concurrent vehicle
Method	: other
Year	: 1981
GLP	: yes
Test substance	: other TS
Method	: The age of the animals was not specified, however, the rats weighed between 240 and 270 grams. Food and water were available to the rats ad libitum except during exposures. The female rats were mated to mature males of the same strain on an as-needed basis. Mating was verified by detection of spermatozoa in the vaginal lavage each morning following overnight cohabitation. Mated females were housed individually. Those rats exposed to DME, and those from the control group, were housed in separate rooms after each daily exposure. DME vapors were metered from a stainless steel cylinder, through a flowmeter into a mixing flask. In the mixing flask, the DME was mixed with 10 L/min air stream prior to entry into the exposure chamber. This mixture was introduced into the top of the exposure chamber where it was further diluted with room air to a total flow of 250 L/min. The exposure chambers were 750 L glass and stainless steel chambers. Chamber atmospheres were quantitatively analyzed for DME by gas chromatography. Body weights and food consumption were measured periodically during gestation. The animals were observed for signs of toxicity and changes in behavior upon arrival, at breeding, and daily from days 6-21 of gestation when the dams were sacrificed. The dams were examined for gross pathologic changes, liver and uterine weights were recorded, and reproductive status was determined. Corpora lutea, implantation sites, live and dead fetuses, resorptions, fetal weight, and the number and position of all live, dead, and resorbed fetuses were recorded. The uterus of each apparently "non-pregnant" rat was stained to detect very early resorptions. All live and dead fetuses were weighed and sexed externally and internally and the live fetuses were examined for external alteration. Approximately one-third of the fetuses were examined for visceral alterations, the heads were removed and underwent a head examination. The above fetuses and all those remaining from each litter were examined for skeletal abnormalities.

5. Toxicity

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Date 10.01.2006

Remark Result

- : Strain: Crl: CD(SD)BR
- : The DME concentrations generated in the exposure chambers were 0, 1250 \pm 50, 5000 \pm 230, and 20,000 \pm 580 ppm for the 0, 1250, 5000, and 20,000 ppm groups, respectively.

The only DME-related effect demonstrated among the dams during exposure was a slight decrease in response to sound at the 20,000 ppm DME level. The response of the 5000 ppm group was equivocal.

Pregnancy ratios were 25/27, 24/27, 27/27, and 25/27 for the 0, 1250, 5000, and 20,000 ppm groups, respectively. A summary of other reproductive outcomes (means/litter) are provided in the tables below. All parameters (except sex ratio) are reported as means/litter.

Concentration (ppm):	0	1250	5000	20,000
Corpora Lutea	16.7	16.3	15.2	15.7
Implantations	14.0	15.3	14.7	14.9
# Resorptions	1.0	1.0	1.0	0.9
Total # Fetuses	13.0	14.3	13.7	14.0
Total # of Live Fetuses	13.0	14.3	13.7	14.0
Mean Fetal Weight (g)	3.8	3.7	3.8	3.7
Sex Ratio (% males)	48.5	48.1	50.1	50.5

DME was not shown to be teratogenic at any level of exposure in this study.

Embryo-fetal toxicity was evident at the 20,000 ppm DME level, which was expressed as decreased fetal body weight (of borderline statistical significance in the 20,000 ppm group) and as an increased incidence of several skeletal variations (partial rib development in the lumbar region and partial or complete doubling of one or more vertebral centra). An increased incidence of one skeletal variation (extra ossification centers in the lumbar area), which was exposure-related, was present in the 5000 ppm DME group. In the 1250 ppm group, the only type of variation with an incidence statistically higher than that of the control group was unossified hyoid bones. This statistically significant increase was isolated in that it occurred only in the lowest exposure group and therefore was not considered an adverse effect of the test compound.

Only one malformed fetus occurred in the 20,000 ppm DME group; it had an umbilical hernia. No malformed fetuses were detected in the 5000 ppm or control group. In the 1250 ppm group, one fetus had multiple malformations, one had no right carotid artery, once had no innominate artery, and in another litter one fetus had no innominate artery.

The following table presents incidence data for the variations discussed above. The results are presented as fetuses/litters.

Concentration (ppm):	0	1250	5000	20,000
Variation				
# fetuses examined for skeletal exams	325/25	343/24	370/27	350/25
Rib - rudimentary	2/1	3/3	7/4	21/11*
Rib - extra	0	0	4/2	4/2
Rib - thickened	0	0	0	2/1
Rib - wavy	1/1	0	0	0

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Rib - extensive wavy	1/1	0	1/1	0
Rib - extra ossification center	19/12	32/15	76/23\$	117/23\$
Centrum - dumbbelled	12/7	14/6	29/13	37/15\$
Centrum - bipartite	5/3	8/6	16/9	13/8
Hyoid - partially ossified	12/7	6/6	9/7	6/6
Hyoid - unossified	2/2	14/8*	5/3	8/5
Hyoid - bipartite	1/1	0	0	0

* = Significantly different from control incidence by Fisher's exact test (p<0.05).

\$ = Significantly different from control incidence by two-tailed Mann-Whitney U test (p<0.05).

The "no-effect" level demonstrated for the conceptus was 1250 ppm DME. The skeletal changes noted were those regarded as being normal variants which signified that the dam was stressed sufficiently to express developmental instability inherent in the species. In comparison to maternal level effects, DME was not demonstrated to represent a unique hazard to the rat conceptus.

Test substance : Dimethyl ether, purity 99.9%
Reliability : (1) valid without restriction
 23.11.2005

(37)

Species : mouse
Sex : female
Strain : Swiss Webster
Route of admin. : inhalation
Exposure period : 1 hour
Frequency of treatm. : days 8, 9, 10, or days 12, 13, 14 of gestation
Duration of test :
Doses : 65,000 ppm
Control group : no data specified
Method : other: see remark
Year :
GLP : no
Test substance : no data

Remark : Pregnant Swiss-Webster mice were anesthetized with 6.5 vol% ether during early or late organogenesis. Animals were anesthetized for 1 hour in a 5.0 liter closed circuit anesthesia apparatus. Litters were delivered by cesarean section on day 19 of gestation, weighed, examined and measured.

Result : Ether anesthesia of pregnant mice caused early and late fetal resorptions and skeletal anomalies but did not alter the incidence of soft tissue anomalies. Thus ether did not show to be highly teratogenic, hypoxia might contribute to the embryotoxicity of ether.

Source : Sodes Paris
 Huels AG Marl
 EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

14.11.2005

(86)

Species : mouse
Sex : female
Strain : Swiss Webster
Route of admin. : inhalation
Exposure period : 60-360 min
Frequency of treatm. : no data
Duration of test : no data
Doses : 65,000 ppm

5. Toxicity

Id 60-29-7

Date 10.01.2006

Control group	:	no data specified
Method	:	other: see remark
Year	:	
GLP	:	no
Test substance	:	no data
Remark	:	<p>In a preliminary study, mice had been exposed to 6.5 vol% diethyl ether for various lengths of time (60-360 min). Twenty-four hours later, the number of dead per group was counted. Fifty percent of the mice died after 100 min anesthesia.</p> <p>Pregnant Swiss Webster mice (number of animals not provided) were then anesthetized for one hour in a 5 liter closed circuit vapor exposure chamber with 6.5 vol% diethyl ether during early or late embryogenesis. Fetuses were delivered by cesarean section one day before normal parturition.</p>
Result	:	<p>Ether anesthesia of pregnant mice during early organogenesis caused a significant incidence of fetal resorptions (14/56) and hydronephrosis (2/26). Anesthesia during early or late organogenesis caused a significant incidence of generalized edema (19/172), missing sternum (10/172), unossified phalanges (9/72), and missing cervical vertebrae (10/72). Anesthesia at either stage did not alter fetal bodyweight or crown-rump length. Length of fetal long bones was decreased by treatment during early organogenesis. Histologic examination of fetal brain, heart, kidney, liver, and skeletal muscle revealed no changes except hepatic parenchymal cell vacuolation.</p>
Source	:	<p>Sodes Paris Huels AG Marl EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)</p>
26.04.1994		(87)
Species	:	other: Chicken (embryo)
Sex	:	no data
Strain	:	other: White Leghorn
Route of admin.	:	other: Ambient gas phase
Exposure period	:	up to 4 days
Frequency of treatm.	:	5 - 6 hour/day
Duration of test	:	
Doses	:	10,000-20,000 ppm
Control group	:	yes
Method	:	other: see remark
Year	:	
GLP	:	no
Test substance	:	no data
Remark	:	<p>Fertile eggs were exposed to ether in glass chambers. Oxygen supply, ether concentration, temperature, humidity were monitored. One-fifth of embryos was opened on the 10th day. Blood concentration of ether in the embryo and yolk was determined. With the others, incubation was continued until the 18th day. The control group was air-exposed. A total of 1058 embryos was studied.</p>
Result	:	<p>Anomalies were observed in brain, eyes, extremities, beak. However, the peak of teratogenesis by ether in this study was at or near the embryo LD50 caused by the stress of this anesthetic concentration. Cellular death from toxicity of the agent is a relatively simple explanation for the teratogenic effect. Therefore, teratogenicity of ether is doubtful.</p>
Source	:	Sodes Paris

5. Toxicity

Id 60-29-7

Date 10.01.2006

26.04.1994

Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

(91)

5.8.3 TOXICITY TO REPRODUCTION, OTHER STUDIES

Type : other
In vitro/in vivo : In vivo
Species : rat
Sex : male/female
Strain : other
Route of admin. : inhalation
Exposure period : 6 hours/day
Frequency of treatm. : 5 days/week (excluding holidays)
Duration of test : 2 years
Doses : 0, 2000, 10,000 and 25,000 ppm
Control group : yes, concurrent vehicle
Method :
Year : 1986
GLP : yes
Test substance : other TS

Method : A 2-year inhalation study was conducted in male and female rats. Terminal sacrifices occurred at 6, 12, 18, and 24 months. Ten rats/sex/group were sacrificed and necropsied at 6, 12, and 18 months and all rats alive at the 2-year time point. All rats underwent both gross and microscopic examinations. Reproductive organs included in the histopathological evaluation included testis, epididymis, prostate, seminal vesicles, cervix, mammary gland, ovary, uterus, and vagina. The testis was weighed.

Remark : Strain: Crl:CD(SD)BR
Result : No compound-related effects on the reproductive organs of either male or female rats were observed. An increase in the incidence of mammary tumors (benign or malignant) was observed in female rats in the 25,000 ppm DME group. The incidence of mammary tumors was considered not to be compound related because the incidences of tumors in the control group were uncharacteristically low in comparison with the control groups incidence in studies previously conducted at Haskell Laboratory.

Test substance : Dimethyl ether, purity 99.98%

Reliability : (2) valid with restrictions

31.05.2005

(38)

5.9 SPECIFIC INVESTIGATIONS

5.10 EXPOSURE EXPERIENCE

Remark : Diethylether has an irritant action on the mucous membrane of the respiratory tract, it stimulates salivation and increases bronchial secretion; laryngeal spasm may occur. It causes vasodilation which may lead to a severe fall in blood pressure, it reduces blood flow to the kidneys and increases capillary bleeding. The bleeding time is unchanged but the prothrombin time may be prolonged. Leucocytosis occurs after ether anesthesia and convulsions occasionally occur in children or young adults under deep ether anesthesia. Recovery is slow from prolonged anesthesia and postoperative vomiting commonly occurs. Acute overdosage of ether is characterized by respiratory failure followed by cardiac arrest.

5. Toxicity

Id 60-29-7

Date 10.01.2006

Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
14.11.2005 (84)

Remark : Diethylether anesthesia caused detectable blood acetaldehyde levels in 15 patients. Ether dose given was not provided, but blood ether levels were within 1,2 and 1,7 g/l in every patient. The average acetaldehyde concentration was 21 µM which approximates the level found after ethanol intake (blood ethanol level was 1 g/l for two hours). No acetaldehyde could be found in patients anaesthetised without ether. The result supports the suggestion of acetaldehyde appearing as an intermediate during ether metabolism.

Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
14.12.1993 (77)

Remark : The minimal alveolar concentration (MAC) to maintain anesthesia in man is 1,92 Vol-% (19200 ppm = 60 mg/l). At this concentration blood shows diethylether values of about 0,7 g/l.

Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
14.12.1993 (20)

Remark : Depending on dose, acute ether inhalation results, in the following clinical signs (no data provided about duration of exposure).
10,000 ppm = 31 mg/l analgesia
30,000 ppm = 93 mg/l consciousness
30.000 - 50,000 ppm = 93-155 mg/l anesthesia
60.000 - 83,000 ppm = 186-257 mg/l cessation of breathing
>103.000 ppm = 319 mg/l lethal damage
Ether anesthesia can result in vomiting at the end of anesthesia, caused by a direct irritation of the gastric mucosa. Isolated cases of centrilobular liver necrosis and fatty degeneration of liver lobular are described, but a typical damage of liver or kidney tissue is not reported. Ether anesthesia can result in a metabolic acidosis followed by hyperglycemia.

Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
26.04.1994 (44)

Remark : In 1904, 3 children were presented who developed tachycardia, pyrexia, and delirium, before dying on the second day after receiving diethyl ether for orthopedic procedures. Autopsy studies performed on the three patients revealed marked fatty hepatic infiltration. None of the children was noted to show signs of an icterus prior to death. The capability of ether to damage liver tissue is questionable.

Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
14.11.2005 (42)

5. Toxicity

Id 60-29-7

Date 10.01.2006

- Remark** : Henderson and Haggard estimated that a man of average weight would absorb a maximum of 1.25 g of ethyl ether, resulting in a blood concentration of 0.018 g/l, when exposed to an atmospheric concentration of 400 ppm (1,24 mg/l). Further details, such as the duration of exposure, were not provided in the ACGIH review of the study.
- Source** : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
14.11.2005 (57)
- Remark** : Frozen abdominal cadaver skin samples were washed with ether for 30 minutes. Before and after treatment skin was analyzed by electron spectroscopy for chemical analyses (ESCA), which allows a valuable in vitro information about elemental and chemical composition of the skin surface to a depth of about 50 Angstroems. ESCA is used to evaluate the removal of skin lipid from epidermis by measuring changes in the skin's atomic percentage of nitrogen. Ether does not extract lipid from the surface of the skin. As a result, ether does not decrease the barrier properties of the skin.
- Source** : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
14.11.2005 (10)
- Remark** : Repeated or prolonged contact to liquid diethyl ether possibly causes dry scaly fissured dermatitis.
- Source** : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
15.12.1993 (28)
- Remark** : Acute eye and mucosal irritation by ether was evaluated. Volunteers (N=10) were exposed to diethylether for 3 to 5 minutes. After exposure, each individual classified the effect of the vapor. Slight nasal irritation was observed at concentrations of 200 ppm (0,62 mg/l).
- Source** : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
15.12.1993 (80)
- Remark** : Minimum lethal dose is given as 273 mg/kg per os for an adult human.
- Source** : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
26.04.1994 (6)
- Remark** : Diethyl ether anesthesia caused detectable blood acetaldehyde levels in 15 patients. Ether dose given was not provided, but blood ether levels were within 1,2 and 1,7 g/l in every patient. The average acetaldehyde concentration was 21 uM which approximates the level found after ethanol intake (blood ethanol level was 1 g/l for two hours). No acetaldehyde could be found in patients anaesthetised without ether. The result supports the suggestion of acetaldehyde appearing as an intermediate during ether metabolism.
- Source** : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
14.12.1993 (77)

5. Toxicity

Id 60-29-7

Date 10.01.2006

Remark : The minimal alveolar concentration (MAC) to maintain anesthesia in man is 1,92 Vol-% (19200 ppm = 60 mg/l). At this concentration blood shows diethyl ether values of about 0,7 g/l.

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (20)

14.11.2005

Remark : Depending on dose, acute ether inhalation results, in the following clinical signs (no data provided about duration of exposure).

10,000 ppm = 31 mg/l analgesia
30,000 ppm = 93 mg/l consciousness
30.000 - 50,000 ppm = 93-155 mg/l anesthesia
60.000 - 83,000 ppm = 186-257 mg/l cessation of breathing
>103.000 ppm = 319 mg/l lethal damage

Ether anesthesia can result in vomiting at the end of anesthesia, caused by a direct irritation of the gastric mucosa. Isolated cases of centrilobular liver necrosis and fatty degeneration of liver lobular are described, but a typical damage of liver or kidney tissue is not reported. Ether anesthesia can result in a metabolic acidosis followed by hyperglycemia.

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (44)

26.04.1994

Remark : In 1904, 3 children were presented who developed tachycardia, pyrexia, and delirium, before dying on the second day after receiving diethyl ether for orthopedic procedures. Autopsy studies performed on the three patients revealed marked fatty hepatic infiltration. None of the children was noted to show signs of an icterus prior to death. The capability of ether to damage liver tissue is questionable.

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (42)

14.11.2005

Remark : Ether inhalation leads to a rapid narcosis starting after signs of preliminary irritation; if ether is given for long enough and in sufficient concentration death takes place from respiratory paralysis. Recovery on removal from exposure to non-lethal concentrations is rapid and there are no apparent cumulative or after-effects. Ether anesthesia is accompanied by acidosis and hyperglycemia. Lower concentrations result in drowsiness, confusion, excitement, dizziness and faintness. After-effects of acute intoxication include nausea, headache, lack of appetite, vomiting, perspiration, mental confusion and irritability. One fatal case is reported, where ether was used in perfumery manufacture as an extracting agent. The subject developed acute mania and died in uraemic convulsions.

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (17)

14.11.2005

Remark : Henderson and Haggard estimated that a man of average weight would

5. Toxicity

Id 60-29-7

Date 10.01.2006

absorb a maximum of 1.25 g of ethyl ether, resulting in a blood concentration of 0.018 g/l, when exposed to an atmospheric concentration of 400 ppm (1,24 mg/l). Further details, such as the duration of exposure, were not provided in the ACGIH review of the study.

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
15.12.1993 (57)

Remark : A wide variation of symptoms is seen following chronic exposure, including occasional dizziness, faintness, loss of appetite and distaste for food, increased thirst (but vomiting when water was taken), nausea, constipation, lassitude, specks before the eyes, numbness in fingers and feet. Nephritis is not frequent but may occur. Some individuals show albuminuria. Ether abuse by drinking leads to "ether habit", chronic effects are inflammation of the respiratory passages, irritability, restlessness, sleeplessness, general debility, headache, and other nervous symptoms, cardiac irregularity and dilatation of blood vessels. Concentrations and/or dosages are not given in this reference.

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
17.02.1997 (17)

Remark : Frozen abdominal cadaver skin samples were washed with ether for 30 minutes. Before and after treatment skin was analyzed by electron spectroscopy for chemical analyses (ESCA), which allows a valuable in vitro information about elemental and chemical composition of the skin surface to a depth of about 50 Angstroems. ESCA is used to evaluate the removal of skin lipid from epidermis by measuring changes in the skin's atomic percentage of nitrogen. Ether does not extract lipid from the surface of the skin. As a result, ether does not decrease the barrier properties of the skin.

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
14.11.2005 (10)

Remark : Repeated or prolonged contact to liquid diethyl ether possibly causes dry scaly fissured dermatitis.

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
15.12.1993 (28)

Remark : Acute eye and mucosal irritation by ether was evaluated. Volunteers (N=10) were exposed to diethyl ether for 3 to 5 minutes. After exposure, each individual classified the effect of the vapor. Slight nasal irritation was observed at concentrations of 200 ppm (0,62 mg/l).

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
14.11.2005 (80)

Remark : Minimum lethal dose is given as 273 mg/kg per os for an adult human.

Source : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

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Date 10.01.2006

26.04.1994

(6)

5.11 ADDITIONAL REMARKS

Type : Behaviour

Remark : NIH mice were exposed to a range of concentrations of ether (1,000-30,000 ppm) in an inhalation chamber and both behavioral and neuroendocrine responses were assessed. When responding was maintained under FI-60s schedules of milk presentation, 30 min exposure to 1,000 ppm ether resulted in minimal behavioral effects, 3,000 - 10,000 ppm increased rates of responding over two-fold and higher concentrations decreased responding almost completely. Five-min exposure to the same range of concentration resulted in concentration-related effects which were smaller than those produced by 30-min exposures. Exposure to a similar range of concentrations in naive mice increased adrenocorticotrophic hormone (ACTH) and corticosterone levels in a time- and concentration-dependent manner. Five-min exposures to 10,000 ppm ether increased levels of ACTH from a baseline of 25.95 pg/ml to 310.5 pg/ml but did not effect corticosterone. Thirty-min exposures to the full range of concentrations of ether increased corticosterone from control levels of 70 ng/l to 418 ng/l at 30,000 ppm, in a concentration dependent manner.

Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

27.04.1994

(48)

Type : Cytotoxicity

Remark : The effect of ether on the division of Chinese hamster fibroblasts in spinner cultures were studied. Ether caused dose dependent inhibition of cell multiplication. ED50 for ether (effective dose, where cell multiplication was reduced to 50% of that of controls, controls were exposed to carrier gas) was 5,97%. Carrier gas was 5% CO2 in air.

Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

27.04.1994

(96)

Type : Neurotoxicity

Remark : Male mice, C57BL/6J and DBA/J2, were repeatedly exposed for 9 seconds (control: air).
Method: Animals had to perform a daily learning trial (escape from shock, six times/day). After familiarization with the test apparatus and a first trial, animals were placed in a jar containing cotton saturated with diethyl ether.
Result: The study revealed that an approximate 9-sec post trial exposure to ether, not resulting in loss of the righting response, can enhance performances of DBA/2J mice. It has no significant effect upon performances of C57/6J mice.

Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

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(116)

Type : other: Embryotoxicity

Remark : Pregnant rabbits were exposed to ether anesthesia. There was observed a significant decrease (>50 %) of the oxygen partial pressure (pO₂) in the fetuses when compared to the pO₂ of the dams. This reduction was probably caused by a decrease of blood pressure. No further data were reported concerning dose or time of exposure.

Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

15.12.1993

(81)

Type : other: Mutation data

Remark : Species: Mouse
Sex: male
Application: 1000 mg/kg, Single dose, i.p.

Method: DNA synthesis inhibition test: By binding covalently to DNA chemical mutagens and carcinogens inhibit replication which can be measured as a decrease of thymidine incorporation into DNA. This DNA synthesis inhibition can be determined in testicular cells of mice. Mice received methyl-14C-thymidine, the following day they received the test substance and subsequently methyl-3H-thymidine. Testes were transferred and homogenized, DNA-content was measured.

Result: False positive, as cytotoxic effect of the anesthetic decreases thymidine incorporation too. When methyl-3H-thymidine was not administered to the animal but to the homogenized testes, no inhibition of DNA synthesis could be observed, the result then was negative.

Source : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

15.12.1993

(89)

Type : other: Mutation data

Remark : Method: P3478 E. coli technique, prescreen for chemical carcinogens: Differential growth inhibition was evaluated as a rapid screening technique for chemical carcinogens.
Test system: E. coli, P3478 (DNA-polymerase-deficient mutant).
Metabolic activation: with and without
Result: negative

Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

15.12.1993

(45)

Type : other: Mutation data

Remark : Method: P3478 E. coli technique, prescreen for chemical carcinogens: Differential growth inhibition was evaluated as a rapid screening technique for chemical carcinogens.
Test system: E. coli, P3478 (DNA-polymerase-deficient mutant).
Metabolic activation: with and without
Result: negative

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- Source** : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
15.12.1993 (45)
- Type** : other: Other relevant data for carcinogenicity
- Remark** : Inhaled ether stimulated tumor growth in mice with subcutaneously or intravenously implanted tumor cells. In the same study, the mitotic index of implanted tumor cells in rats was not affected by administration of an unspecified concentration of diethyl ether.
- Source** : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
15.12.1993 (46)
- Type** : other: Other relevant data for carcinogenicity
- Remark** : Diethyl ether and disulfiram dissolved in diethyl ether was administered to 8- and 9-day embryos (inbred CH3 mice) in vitro in concentrations of 0.285 mg/ml and 2.85 mg/ml. Apart from a reduction in somite counts, ether in these concentrations caused no adverse effects on morphological development in 8- or 9-day embryos. DNA synthesis was inhibited at a concentration of 2.85 mg/ml in 9-day embryos.
- Source** : Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
15.12.1993 (99)
- Type** : other: Toxicokinetics/Metabolism
- Remark** : Exposure of 20 h fasted male Wistar albino rats, ca. 250 g, to ether anesthesia for 6 min (dose level of approx. 5 g/kg) resulted in increased exhalation of alkanes, and indication of lipid peroxidation in vivo. Total cytochromes P-450 of liver and kidney were decreased to 25-30 % of control values, but were restored to normal levels 2 h later. Cytochrome P450 I (EROD activity) was decreased to 35-44 % of control values and was restored to 80 % of normal levels 2 h later. Diethyl ether is known to be metabolized by cytochrome P450IIE1 which is induced by fasting and by diethyl ether, and is possibly involved in the observed radical production, lipid peroxidation, and loss of cytochromes P-450. The effect of ether seen in this study could readily explain the hepatic necrosis seen in fatalities following prolonged ether anesthesia.
- Source** : Sodes Paris
Huels AG Marl
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
15.12.1993 (75)
- Type** : other: Toxicokinetics/Metabolism
- Remark** : The (acute) effect of diethyl ether anesthesia on in vivo hepatic protein synthesis was tested in male Wistar rats. Protein synthesis was measured by an isotope technique. It was shown that usual anesthetic levels of diethyl ether reduced the rate of synthesis of liver proteins to 80 % compared to a group receiving no anesthesia. The synthesis /secretion of plasma proteins was much more inhibited, to approximately 20-30 %, compared to animals either receiving no anesthesia or pentobarbital. No further data were given

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Source : about number of animals or dosages.
: Sodes Paris
14.11.2005 EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (12)

Type : other: Toxicokinetics/Metabolism

Remark : Eighteen male Sprague Dawley rats were instrumented with microspheres. Cardiac output and blood flow distribution were determined at five different periods: before ether anesthesia; at a surgical level of ether anesthesia; and 20 min, 1 hr, or 3 hr after cessation of anesthesia. Ether anesthesia initially decreased arterial pressure, increased cardiac index, and decreased total peripheral resistance. The residual effects of ether included progressive increases in arterial blood pressure and an increase in total peripheral resistance index. Cardiac index was returned to normal 1 hr after termination of anesthesia. Blood flow to the brain and heart increased during anesthesia and was significantly elevated 1 hr later. Other organs, including kidney, spleen, and intestine showed a decrease in blood flow during anesthesia, which persisted for at least 20 min. Thus ether anesthesia produced acute and residual disturbances in hemodynamics and blood flow distribution. No further data were given about the ether dose administered.

Source : Sodes Paris
15.12.1993 EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (93)

Type : other: Toxicokinetics/Metabolism

Remark : Rats were exposed to an ether concentration of 10 % v/v (= 310 mg/l) for 5-60 min. Ether concentration was determined in omental and renal fat and given as mg ether per 1 g of tissue wet weight. A few minutes after administration the concentration of ether in fatty tissue was the same as in blood, after 15 minutes it was considerably higher in fat than in blood. The maximum concentration in fatty tissue (about 3 mg/g) was reached after 0.5-1 h. The elimination of ether from fatty tissues in rats did not begin immediately after the end of ether administration, but only when the concentration in the blood had become relatively low. It was practically finished after about 8 h. 24 h after a 1 h exposure, ether concentration in fatty tissue was 0.12 mg/g and 0.03 in blood, respectively.

Source : Sodes Paris
15.12.1993 EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (41)

Type : other: Toxicokinetics/Metabolism

Remark : Dogs were exposed to ether in a closed circuit system. Ether concentration was determined in arterial and venous blood by infrared spectrometry. In case of tissue determination of ether, infrared spectroscopy, mass spectroscopy and roentgenographic fluorescence spectroscopy were used. Concentration of ether in selected tissues after 2.5 h of ether anesthesia (Number of animals: 8; ether dose given not provided):
arterial blood 1.025 mg/g
brain 1.140 mg/g
adrenal 1.945 mg/g

- fat 6.700 mg/g
skel. muscle 0.853 mg/g
liver 0.940 mg/g
kidney 2.420 mg/g
- Source** : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (26)
15.12.1993
- Type** : other: Toxicokinetics/Metabolism
- Remark** : Dogs were exposed to ether anesthesia for a duration of 10 minutes or 2.5 hours. Ether concentration in anesthesial induction period was 10-20 % in oxygen (= concentration of 100,000-200,000 ppm or 308-616 mg/l). At the end of exposure, ether concentration was determined in various areas of the brain and in blood. Ranges of arterial blood concentration: from 0.74 to 1.31 mg/g for 10 minutes of anesthesia, and 0.966 - 1.464 mg/g for 2.5 h of anesthesia. The ratio of brain to blood concentration at 10 minutes was from 0.7 to 1.8.
- Source** : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (25)
15.12.1993
- Type** : other: Toxicokinetics/Metabolism
- Remark** : Rats were dosed by intraperitoneal injection with isotopically labeled ether at a dose level of 357 mg/kg. Animals were placed in all-glass metabolism cage to allow the recovery of the expired gases and separate collection of urine and feces. The animals remained in these containers for periods up to 96 h. Rats were narcotized for periods up to 2 h.
The total radioactivity in CO₂ and urine collected in a 24 hour period was 4 % and 2 %, respectively, of the amount injected. The authors summarized that ether is not inert but undergoes a biotransformation. They suggested that ether is cleaved by O-dealkylation, which occurs under catalysis of an enzyme found in microsomes.
- Source** : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (109)
15.12.1993
- Type** : other: Toxicokinetics/Metabolism
- Remark** : Isotopically labeled ether was administered to NMRI mice by inhalation. Animals were exposed to a concentration of 80,000 ppm (246 mg/l), exposure duration of 15 minutes or 2 h, respectively. Uptake and metabolism of ether were studied with whole-body autoradiography. Increased relative concentration of radioactivity developed in liver and kidney, reflecting the accumulation of nonvolatile metabolites. At two hours the measured nonvolatile metabolites accounted for 3.6 % of the administered radioactivity. Further investigation of an extract of liver showed the presence of four nonvolatile metabolites. The authors suggest the following mechanism for ether metabolism:
- CH₃-CH₂-O-CH₂-CH₃ -> CH₃-CH₂-O-CH(OH)-CH₃
-> CH₃-CH₂OH + CH₃-CHO
diethyl ether -> ethanol + acetaldehyde

- However, it was considered that there were additional pathways, as the investigation of liver extract indicated the presence of a glucuronide of ether.
- Source** : Sodes Paris
15.12.1993 EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (27)
- Type** : other: Toxicokinetics/Metabolism
- Remark** : Method: Isotopically labeled ether was administered to NMRI mice by inhalation. The animals were sacrificed 2 h after anesthesia. Livers were investigated for nonvolatile metabolites of ether by measuring the radioactivity of the liver extract. Approximately 1 % of the administered radioactivity was recovered in the extract. The extracted metabolites were then separated by thin layer chromatography.
Result: A portion of diethyl ether administered by inhalation was rapidly transformed into fatty acids (palmitic, stearic and oleic acids), cholesterol, mono-, di- and triglycerides. The authors suggest that diethyl ether is transformed to acetate which enters the common metabolic pool and is subsequently degraded to CO₂.
- Source** : Sodes Paris
15.12.1993 EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (50)
- Type** : other: Toxicokinetics/Metabolism
- Remark** : Rats were treated with phenobarbital (80 mg/kg i.p.) for 3 days prior to sacrifice (24 h after final injection). Hepatic microsomes were prepared. The incubations consisted of NADP, MgCl₂, glucose-6-phosphate, EDTA, glucose-6-phosphate dehydrogenase, microsomal protein, and diethyl ether in potassium phosphate buffer. Acetaldehyde formation was determined by photometry. The experiment indicated that diethyl ether was metabolized to acetaldehyde by microsomes and that this reaction was linear through 20 min. This reaction required NADPH and was inhibited by both carbon monoxide and antibody to rat liver cytochrome P-450. The authors suggest that microsomal metabolism of diethyl ether is catalysed by a cytochrome P-450-containing mono-oxygenase system.
- Source** : Sodes Paris
15.12.1993 EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (23)
- Type** : other: Toxicokinetics/Metabolism
- Remark** : Isolated rat liver parenchymal cells incubated with anesthetic concentrations of diethyl ether were shown to produce acetaldehyde and ethanol in a dose dependent manner. The acetaldehyde and ethanol production from ether was stimulated in hepatocytes derived from phenobarbital treated rats and could be only partially inhibited by 4-methyl pyrazole. The study results support the suggestions that diethyl ether is metabolized by an inducible microsomal enzyme system which cleaves diethyl ether in a reaction analogous to the known O-dealkylation reactions.
- Source** : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

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Type : other: Toxicokinetics/Metabolism

Remark : The ethanol disappearance rate was determined in fed rats given 20-40 mMol ethanol and anesthetized with pentobarbital (control group) and diethyl ether. Rats anesthetized with diethyl ether (blood levels of 9-13 mM) revealed a 52% inhibition of ethanol disappearance when compared to control. This observation indicated that the site of inhibition could be referred to the cytosolic enzyme alcohol dehydrogenase.

Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

15.12.1993

(82)

Type : other: Toxicokinetics/Metabolism

Remark : The effect of diethyl ether on ethanol metabolism was studied in isolated rat hepatocytes and ether was found to inhibit ethanol oxidation in a dose-dependent manner. At ethanol concentrations of approximately 30 mM, diethyl ether inhibited ethanol oxidation by approximately 58, 40, and 20% at ether concentrations of 30, 20, and 10 mM, respectively.

Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

14.11.2005

(9)

Type : other: Toxicokinetics/Metabolism

Remark : Method: Diethyl ether was administered by inhalation to Sprague Dawley rats at a concentration of 16,000 ppm (49.6 mg/l). Exposure time was 7 h/day for five days. A control group was exposed to air. Animals were sacrificed and liver microsomes were prepared. Microsomal cytochrome P-450, NADPH cytochrome c reductase and cytochrome b5 were determined. Liver triglycerides were determined. A histologic evaluation of liver was performed.
Result: The study showed that diethyl ether increases microsomal cytochrome P-450, NADPH cytochrome c reductase, cytochrome b5 and microsomal protein. No change in hepatic architecture was observed.

Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

15.12.1993

(16)

Type : other: Toxicokinetics/Metabolism

Remark : Method: Mice were treated i.p. with sodium phenobarbital for 3 days and with beta-naphthoflavone on the 2nd day. Each day, immediately prior to treatment, the mice were exposed to an anesthetic ether atmosphere (about 1 min). Controls were treated the same way but without any ether anesthesia.
Animals were killed by cervical dislocation, part of the ether dosed animals was killed by an over-dose of ether (inhalation about 3-5 min). Livers were homogenized. Aminopyrine demethylase, p-nitroanisole demethylase and protein were determined. The enzyme stability in the conditions of the liver microsomal assay was followed by determining the activities.

Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)

15.12.1993

(11)

- Type** : other: Toxicokinetics/Metabolism
- Remark** : Method: Male Sprague Dawley rats were pretreated with several agents to induce liver metabolism, i.a. phenobarbital, butylated hydroxytoluene, acetone, ethanol, and isoflurane. Animals were then exposed to an atmosphere saturated with diethyl ether until loss of righting reflex. Ether treatment was repeated three times or five times daily for 3 days. In a second experiment, liver induction was performed by pretreatment with ether in the described way. Animals were sacrificed the fourth day. Liver microsomes were prepared. Demethylase activities, ether deethylase activity and O-dealkylation were estimated and immunoblot analysis was performed.
Result: Microsomal oxidation of ether to acetaldehyde was elevated 1.5- to 2-fold by pretreatment with ether when compared to control. Ether also induced N-nitrosodimethylamine demethylase b up to 2-fold and O-dealkylation by up to 10-fold. These trends agreed with the result of the immunoblot experiment in which ether was an inducer of the P-450 isoenzyme IIE1, but a stronger inducer of IIB1. N-nitrosodimethylamine, as well as common inhibitors of IIE1 such as hexane strongly inhibited deethylation.
- Source** : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
14.11.2005 (13)
- Type** : other: Toxicokinetics/Metabolism
- Remark** : Rats were pretreated with 1) microsomal enzyme inducers, 2) inhibitors of microsomal enzymes, 3) hepatotoxins, 4) commonly used anesthetics, e.g. diethyl ether. Ether was administered by inhalation to induce and maintain narcosis for 10 min before sacrifice. Animals were sacrificed and livers prepared. Hepatic UDPGA (UDP-glucuronosyltransferase) content was decreased to 5 % of the control after exposure to ether. This result indicated that ether was able to influence the rate of glucuronidation to a high extent.
- Source** : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
14.11.2005 (112)
- Type** : other: Toxicokinetics/Metabolism
- Remark** : Rats were fitted with bile duct and jugular vein catheters while anesthetized with diethyl ether. As anesthetize abated, bile was collected for the next 5 h and analyzed for flow rate, total bilirubin excretion, and bilirubin glucuronide composition. The HPLC method used allowed direct analysis of bile without derivation or extraction. Ether anesthesia was associated with a reversible suppression of diglucuronide formation and total bilirubin excretion, with reciprocal monoglucuronide changes. These results supported the hypothesis that alterations in UDP-glucuronic acid concentration were capable of influencing rates of hepatic glucuronide formation.
- Source** : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
15.12.1993 (49)
- Type** : other: Toxicokinetics/Metabolism

- Remark** : Acetaminophen (Paracetamol) is toxified by cytochromes P-450 to a hepatotoxic reactive metabolite. Brief general anesthesia with diethyl ether has been shown to inhibit both the toxifying cytochromes P-450 and enzymatic glucuronidation, the latter constituting up to 60 % of acetaminophen elimination via a nontoxic pathway. Thus ether could potentially produce a temporally differentiated inhibition of bioactivating and detoxifying pathways, resulting in an enhancement of acetaminophen toxicity if the balance favored bioactivation. To evaluate this possibility, male NIH mice were treated with acetaminophen at different times after 5 min of anesthesia with ether. Ether produced a 40-fold enhancement in acetaminophen hepatotoxicity as determined by the increase of plasma GPT (= ALT, alanine amino transferase) concentrations. These results showed that glucuronidation was inhibited by ether anesthesia.
- Source** : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
15.12.1993 (113)
- Type** : other: Toxicokinetics/Metabolism
- Remark** : Method: Ether was administered over 5 min to groups of 6 male CD-1 mice housed in an inhalation chamber till loss of righting reflex. Acetaminophen (= Paracetamol; 300 mg/kg) was injected i.p. at 2, 6, or 10 h after ether anesthesia. Hepatocellular damage was assessed by determining concentration of plasma GPT (= ALT, alanine amino transferase). Animals were sacrificed and livers prepared.
- Result: Brief ether anesthesia resulted in 1) an increase of covalent binding of acetaminophen to hepatocellular protein, 2) a delayed decrease of hepatic activity of glucuronyl transferase, 3) a delayed decrease of hepatic activity of GSH (glutathione) sulfotransferase, 4) an initially reduced hepatic content but an unchanged activity of cytochrome P-450, 5) a delayed reduction of hepatic GSH contents, 6) an increase of plasma GPT indicating liver damage. This biochemical mechanism of this potentiation of hepatotoxicity was supposed to be due to delayed, complex effects of ether upon multiple enzymatic pathways of acetaminophen elimination and detoxification.
- Source** : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA)
23.11.2005 (100)
- Type** : other: Toxicokinetics/Metabolism
- Remark** : Method: Ether was administered over 5 min to male CD-1 mice housed in an inhalation chamber till loss of righting reflex. Acetaminophen (APAP = Paracetamol; 300 mg/kg) was injected i.p. at various times after ether anesthesia. Hepatocellular damage was assessed by determining concentration of plasma GPT (= ALT, alanine amino transferase). Animals were sacrificed and livers prepared. The in vitro activities of enzymes responsible for 1) elimination (= glucuronyl transferase), 2) detoxification (= glutathione sulfotransferase) and 3) bioactivation / toxification (= cytochromes P-450) of APAP were estimated. The in vivo elimination of APAP and its metabolites was determined in plasma and urine. The covalent binding of APAP to hepatocellular protein in vivo was determined.

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Result: Ether could initially inhibit both the elimination pathway and the toxifying pathway. It was shown that the toxifying pathway recovers first and causes by this way enhanced hepatotoxicity.

Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (114)
15.12.1993

Type : other: Toxicokinetics/Metabolism

Remark : The effects of two anesthetic procedures 1) continuous administration of ether throughout the periods of drug infusion (antipyrine and paracetamol) and blood sampling and 2) brief ether administration before drug infusion were examined. Ether was administered to rats till loss of righting reflex for 5 min or several hours. Continuous ether caused substantial reductions in the elimination rates of antipyrine and paracetamol. Brief ether anesthesia had no effect on antipyrine kinetics, but caused a decrease in total clearance of paracetamol. The rates of distribution and redistribution were unchanged by ether. This suggested that ether interfered with the hepatic conjugation of paracetamol and might interfere with the hepatic oxidation of antipyrine.

Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (67)
15.12.1993

Type : other: Toxicokinetics/Metabolism

Remark : The response of two different pathways of paracetamol metabolism to diethyl ether was examined. The elimination of paracetamol and the formation of paracetamol sulphate and glucuronide were measured in suspensions of isolated rat hepatocytes from fasted and fed animals over 1 h in the absence and presence of diethyl ether (30 mmol/l). Approximately 90 % of the paracetamol elimination was by sulphation and nearly 10 % by glucuronidation both in the controls and in the presence of ether. The overall disposition of paracetamol and the formation of sulphate were both reduced by about 50 % in the presence of ether compared to the controls while the formation of glucuronide was reduced by 70 %. The results were not influenced by the nutritional state of the animals before sacrifice. It is concluded that the inhibitory effect of ether on total paracetamol metabolism was mainly caused by reduced sulphation. Since microsomal glucuronidation was also inhibited by ether, both cytosolic and microsomal enzyme systems were sensitive to ether.

Source : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (8)
15.12.1993

Type : other: Toxicokinetics/Metabolism

Remark : The effect of diethyl ether on rat-liver microsomal glucuronyltransferase activity was examined in vitro. Diethyl ether depressed this reaction in a dose-related, noncompetitive manner. Glucuronyltransferase activity was studied by measuring the rate of glucuronide conjugation of p-nitrophenol in the presence of various concentrations of

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- ether in atmosphere (15, 20, 30 mM). Inhibition occurred when UDPGA (uridine diphosphoglucuronic acid) was used as a glucuronic acid donor and to a higher extent when the UDPGA-generating system (UDPG + NAD) was employed.
- Source** : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (15)
15.12.1993
- Type** : other: Toxicokinetics/Metabolism
- Remark** : In vitro experiments with rat liver microsomes (p-nitroanisole-demethylation) and in vivo experiments with rats (tritium release from 3H-mestranol due to demethylation) suggested an inhibition of the metabolism of certain drugs by competition for the binding site of cytochrome P-450 under anesthesia with diethyl ether.
- Source** : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (56)
15.12.1993
- Type** : other: Toxicokinetics/Metabolism
- Remark** : The effect of ether stress on mixed function oxidase activity in vivo was studied using the aminopyrine-¹⁴CO₂ exhalations rate method. Ether was administered to rats by inhalation for 6 h. Animals were transferred to a metabolism cage where a constant subanesthetic concentration of ether was maintained. Ether exposure did not produce any consistent effect on drug metabolizing status of the rat.
- Source** : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (76)
15.12.1993
- Type** : other: Toxicokinetics/Metabolism
- Remark** : The in vitro effect of diethyl ether on the Michaelis constant (K_m) and maximal velocity (V_{max}) of microsomal aniline hydroxylase and aminopyrine demethylase was determined. The microsomes were obtained from rats pretreated with phenobarbital or 3-methylcholanthrene as well as from untreated rats. Diethyl ether inhibited aniline hydroxylase lowering the V_{max} and aminopyrine demethylase by increasing K_m.
- Source** : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (73)
15.12.1993
- Type** : other: Toxicokinetics/Metabolism
- Remark** : The effect of ether as a possible antagonist of mediator-effected bronchoconstriction was tested in eight anesthetized, paralyzed and mechanically ventilated baboons. Ether administration was performed intravenously, a 13 minutes infusion to give approximately 1:3 MAC (minimum alveolar anesthetic concentration, baboons were assumed to have the same MAC as human patients). Ether had no effect on bronchoconstriction caused by acetylcholine, histamine, or phenylephrine administered by the same route.
- Source** : Sodes Paris
EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (79)
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Remark	: Dogs were exposed to an atmosphere of diethyl ether. About 87 % of the inspired ether was excreted unchanged in the expired air at the end of the experiment. Traces of ether were found in the urine (2 %, concentration approximately equal to that of blood passing through the kidneys). A slight accumulation of ether in fatty tissue was observed.
Source	: Sodes Paris
14.11.2005	EUROPEAN COMMISSION - European Chemicals Bureau Ispra (VA) (52)

6.1 ANALYTICAL METHODS

6.2 DETECTION AND IDENTIFICATION

7.1 FUNCTION

7.2 EFFECTS ON ORGANISMS TO BE CONTROLLED

7.3 ORGANISMS TO BE PROTECTED

7.4 USER

7.5 RESISTANCE

8.1 METHODS HANDLING AND STORING

8.2 FIRE GUIDANCE

8.3 EMERGENCY MEASURES

8.4 POSSIB. OF RENDERING SUBST. HARMLESS

8.5 WASTE MANAGEMENT

8.6 SIDE-EFFECTS DETECTION

8.7 SUBSTANCE REGISTERED AS DANGEROUS FOR GROUND WATER

8.8 REACTIVITY TOWARDS CONTAINER MATERIAL

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10.1 END POINT SUMMARY

10.2 HAZARD SUMMARY

10.3 RISK ASSESSMENT

**EPA Comments on Chemical RTK HPV Challenge Submission: Diethyl Ether
Summary of EPA Comments**

The sponsor, the Diethyl Ether Producers Association (DEPA), submitted a test plan and robust summaries to EPA for Diethyl ether (1,1'-Oxybisethane; CAS No. 60-29-7) dated December 30, 2003. EPA posted the submission on the ChemRTK HPV Challenge Web site on February 25, 2004.

EPA has reviewed this submission and has reached the following conclusions:

- 1 Physicochemical Properties. The data provided by the submitter for these endpoints are adequate for the purposes of the HPV Challenge Program.
- 2 Environmental Fate. The submitter needs to provide 28 day biodegradation data for this chemical.
- 3 Health Effects. EPA agrees with the submitter's plan to conduct a combined repeated-dose/reproductive/developmental toxicity screening test and a chromosomal aberration assay following OECD guidelines. The submitter needs to address deficiencies in the robust summaries.
- 4 Ecological Effects. EPA agrees with the submitter's plan to test algae, and with the adequacy of the data submitted on fish and invertebrates.

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EPA Comments on the Diethyl Ether Challenge Submission

Test Plan

Physicochemical Properties (melting point, boiling point, vapor pressure, water solubility, and partition coefficient)

The data provided by the submitter for these endpoints are adequate for the purposes of the HPV Challenge Program.

RESPONSE: Acknowledged.

Environmental Fate (photodegradation, stability in water, biodegradation, fugacity)

The data provided by the submitter for photodegradation, stability in water and fugacity are adequate for the purposes of the HPV Challenge Program.

RESPONSE: Acknowledged.

Biodegradation. The submitter provided one biodegradation study similar to OECD Guideline 301C, in which it indicates that no biodegradation was observed after 10 days. This information is not sufficient to conclude that this chemical is not readily biodegradable. The submitter needs to

provide 28-day test data following OECD Guideline 301 or from reliable published literature sources. EPA found biodegradation information for this chemical in the website for the Chemicals Evaluation and Research Institute, Japan (CERIJ), at: http://qsar.cerij.or.jp/cgi-bin/QSAR/e_r_text_query.cgi. EPA recommends that the submitter add this information into its biodegradation robust summary.

RESPONSE: See revised Test Plan incorporating additional information for Diethyl Ether. As requested by the Agency, the information from CERIJ, Japan have been added to the IUCLID and revised Test Plan.

Health Effects (acute toxicity, repeated-dose toxicity, genetic toxicity, and reproductive/developmental toxicity)

EPA agrees that adequate data are available for acute toxicity for the purposes of the HPV Challenge Program but reserves judgement on the data submitted for gene mutations pending receipt of additional information. EPA agrees with the submitter's plan to conduct a combined repeated-dose/reproductive/developmental toxicity screening test and a chromosomal aberration assay following OECD guidelines (OECD TGs 422 and 473, respectively).

RESPONSE: The Diethyl Ether Producers Association has reevaluated the proposed testing in light of the availability of data for Dimethyl Ether. Data for Dimethyl Ether have been incorporated into the dataset for DEE and a rationale for using DME data has been included in the Revised Final Test Plan for DEE.

Ecological Effects

EPA agrees with the submitter's plan to test in algae. The data submitted on fish and invertebrates are adequate.

RESPONSE: The Diethyl Ether Producers Association has reevaluated the proposed testing in light of the availability of data for Dimethyl Ether. Data for Dimethyl Ether have been incorporated into the dataset for DEE and a rationale for using DME data has been included in the Revised Final Test Plan for DEE.

Specific Comments on the Robust Summaries

Health Effects

Acute toxicity. A robust summary for an acute oral toxicity study in rats omitted the identity and purity of the test substance.

RESPONSE: No data were provided in the reported study on the purity of the test substance. Because DEE is normally produced at a very high purity, it is presumed that a material of high purity (> 99%) was used.

Genetic toxicity (gene mutations). Details missing in a robust summary for an Ames test include the purity of the test material, concentrations tested, cytotoxic concentration, number of colonies counted per concentration, information on positive and negative controls and the statistical methods.

RESPONSE: No data for these parameters was provided in the publication that tested a large number of chemicals in the Ames assay.

Followup Activity

EPA requests that the submitter advise the Agency within 60 days of any modifications to its submission.